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# APPETITE SUPPRESSANT ACTIVITY

THIS INVENTION relates to steroidal glycosides, to compositions containing such steroidal glycosides and to a new use for these steroidal glycosides and the compositions containing them. The invention further relates to a method of extracting and isolating these steroidal glycosides from plant material, to a method of synthetically producing these steroidal glycosides, and to the products of such an extraction and such a synthesis process.

In a particular application, the invention relates to an appetite suppressant agent, to a process for synthetically producing the appetite suppressant agent, to a process for extracting the appetite suppressant agent from plant material, to an appetite suppressant composition containing the appetite suppressant agent, and to a method of suppressing an appetite.

According to the invention, there is provided a process for preparing an extract of a plant of the genus Trichocaulon or of the genus Hoodia, the extract comprising an appetite suppressant agent, the process including the steps of treating collected plant material with a solvent to extract a fraction having appetite suppressant activity, separating the extraction solution from the rest of the plant material, removing the solvent from the extraction solution and recovering the extract. The extract so recovered may be further purified, eg by way of suitable solvent extraction procedures.

The invention also provides a plant extract made of plants of the group comprising the genus *Trichocaulon* and the genus *Hoodia* and having appetite suppressant activity.

The extract may be prepared from plant material such as the stems and roots of said plants of the genus

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genus Hoodia. Trichocaulon or of the The Trichocaulon and the genus Hoodia include succulent plants growing in arid regions such as are found in Southern In one application of the invention, the active appetite suppressant extract is obtained from the species Trichocaulon piliferum. The species Trichocaulon officinale may also be used to provide an active appetite suppressant extract. In another application of invention, the active appetite suppressant extract may be obtained from the species Hoodia currorii, Hoodia gordonii or Hoodia lugardii. Bioassays conducted by the Applicant on rats have indicated that certain of the extracts possess appetite suppressant activity.

The plant material may be homogenised in the presence of a suitable solvent, for example, a methanol/methylene chloride solvent, by means of a device such as a Waring blender. The extraction solution may then be separated from the residual plant material by an appropriate separation procedure such as, for example, filtration or centrifugation. The solvent may be removed by means of the rotary evaporator, preferably in a water bath at a temperature of 60°C. The separated crude extract may then be further extracted with methylene chloride and water before being separated into a methylene chloride extract The methylene chloride extract may and a water extract. have the solvent removed preferably by means of evaporation on a rotary evaporator and the resultant extract may be further purified by way of a methanol/hexane extraction. methanol/hexane extraction product may then separated to yield a methanol extract and a hexane extract. The methanol extract may be evaporated to remove the solvent in order to yield a partially purified active extract.

The partially purified active extract may be dissolved in methanol, and may be further fractionated by column chromatography, employing silica gel as an adsorption į

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medium and a chloroform/30% methanol mixture as an eluent. A plurality of different fractions may be obtained, and each may be evaluated, by suitable bioassaying procedures, to determine the appetite suppressant activity thereof.

A fraction having appetite suppressant activity may preferably be further fractionated such as by column chromatography using silica gel as an adsorption medium and a 9:1 chloroform:methanol solvent, and the resultant subfractions bioassayed for their appetite suppressant activity. A sub-fraction displaying appetite suppressant activity may, if desired, be further fractionated and purified, conveniently using a column chromatographic procedure with silica gel as the adsorption medium and a 9:1 ethylacetate:hexane solvent. The resultant purified fractions may again be evaluated by suitable bioassay procedures for their appetite suppressant activity.

The Applicant has found that at least one such purified fraction has good appetite suppressant activity, and the active principle in the fraction was identified by conventional chemical techniques including nuclear magnetic resonance, and was found to be a compound of the structural formula

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In accordance with S.I. nomenclature, the active principle (1) is the compound 3-0-[- $\beta$ -D-thevetopyranosyl-(1 $\rightarrow$ 4)- $\beta$ -D-cymaropyranosyl-(1 $\rightarrow$ 4)- $\beta$ -D-cymaropyranosyl]-12 $\beta$ -0-tigloyloxy-14-hydroxy-14 $\beta$ -pregn-50-en-20-one (C<sub>47</sub>H<sub>74</sub>O<sub>15</sub> M\*878).

According to another aspect of the invention, there is provided a process for preparing an extract of a plant of the genus *Trichocaulon* or of the genus *Hoodia*, the extract comprising an appetite suppressant agent, the process including the steps of pressing collected plant material to separate sap from solid plant material and recovering the sap free of the solid plant material to form the extract.

The extract may be dried to remove moisture, e.g. by spray-drying, freeze-drying or vacuum drying, to form a free-flowing powder.

The invention extends to a composition having appetite suppressant activity comprising an extract as described above.

The composition may be admixed with a pharmaceutical excipient, diluent or carrier and optionally it is prepared in unit dosage form.

The invention also extends to the use of an extract as described above in the manufacture of a medicament having appetite suppressant activity, to an extract as described above for use as a medicament having appetite suppressant activity, and to a method of suppressing an appetite by administering to a human or animal an effective dosage of a composition as described above.

Compound (1) is a novel compound and the invention extends to compound (1) and certain analogues or derivatives of this steroidal trisaccharide having appetite suppressant properties. The molecules chosen as the

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analogues or derivatives are intended to affect the properties of the steroidal trisaccharide with the aim of increasing the activity of the active ingredient. The following effects were taken into consideration when the analogues were chosen:

### (i) Hydrophobic interactions and lipophilicity

Functional group modifications of the active molecule intended to change the hydrophobicity lipophilicity ο£ the molecule. Increased lipophilicity has been shown to correlate with increased biological activity, poorer aqueous solubility, increased detergency/cell increased storage in tissues, more rapid metabolism and elimination, increased plasma protein binding and faster rate of onset of action.

#### (ii) Electronic properties and ionization constants

Functional group modification of the molecule is also intended to change the acidity and basicity which would have a major role in controlling the transport of the compound to its site of action and the binding at this target site.

#### (iii) Hydrogen bonding

Functional group modifications of carboxyl and carbonyl groups in the active molecule are intended to change the interactions between the proteins in biological systems and the chemically modified functional groups.

#### (iv) Steric parameters

The purpose of changing the steric features of the molecule is to increase binding to its receptor and

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thus increase its biological activity.

The following chemical modifications to the molecule are intended to affect the hydrophobicity and lipophilicity electronic properties, hydrogen bonding and steric parameters on the molecule:

- a) Chemical modification of the C-12 group and ester functionality;
- b) Chemical modification of the 5,6-double bond, e.g. hydrogenation and migration;
- 10 c) Chemical modification of the C-20 carbonyl and C-17 acetyl group;
  - d) Chemical modification of the "D" ring of the steroid or aglycone ring;
- e) Modification of the carbohydrates of the trisaccharide moiety.

Accordingly, the invention provides a compound having the general structural formula

(2)

in which R = alkyl;

 $R_1 = H$ , alkyl, tigloyl, benzoyl, or any other organic ester group;

 $R_2$  = H, or one or more 6-deoxy carbohydrates, or one or more 2,6-dideoxy carbohydrates, or glucose molecules, or combinations thereof;

and in which the broken lines indicate the optional presence of a further bond between C4-C5 or C5-C6.

The invention also provides a compound as described above wherein there is a further bond between C5 - C6, R = methyl,  $R_1$  = tigloyl,  $R_2$  = 3-0-[- $\beta$ -D-thevetopyranosyl-(1 $\rightarrow$ 4)- $\beta$ -D-cymaropyranosyl-(1 $\rightarrow$ 4)- $\beta$ -D-cymaropyranosyl] and having the structural formula.

Further active analogues or derivatives of the appetite suppressant compound (1) in accordance with the invention are compounds having the following structural formulae:

in which R = alkyl; and

R<sub>1</sub> = H, or benzoyl, or tigloyl, or any other
organic ester group

(5)

in which R = alkyl; and  $R_1 = H$ , or tigloyl, or benzoyl, or any other

#### organic ester group

(6)

in which R = alkyl; and  $R_1 = H,$  or tigloyl, or benzoyl, or any other organic ester group

5 in which R = alkyl;

 $R_1$  = H, or tigloyl, or benzoyl, or any other organic ester group.

(8)

in which R = alkyl; and

 $R_1 = H$ , alkyl, tigloyl, benzoyl, or any other organic ester group;

 $R_2$  = H, or one or more 6-deoxy carbohydrates, or one or more 2,6-dideoxy carbohydrates, or glucose molecules, or combinations thereof;

and in which the broken lines indicate the optional presence of a further bond between C4 - C5 or C5 - C6.

$$R_{2}$$
 $OR_{1}$ 
 $OH$ 
 $OH$ 
 $OH$ 
 $OH$ 
 $OH$ 

R = alkyl; andin which 10

 $R_1 = H$ , alkyl, tigloyl, benzoyl, or any other organic ester group;

 $R_2$  = H, or one or more 6-deoxy carbohydrates, or one or more 2,6-dideoxy carbohydrates, or glucose

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molecules, or combinations thereof; and in which the broken lines indicate the presence of a further bond between C4 - C5 or C5 - C6.

(10)

in which R = alkyl; and

 $R_1$  = H, alkyl, tigloyl, benzoyl, or any other organic ester group;

 $R_2$  = H, or one or more 6-deoxy carbohydrates, or one or more 2,6-dideoxy carbohydrates, or glucose molecules, or combinations thereof;

and in which the broken lines indicate the optional presence of a further bond between C4 - C5 or C5 - C6.

$$R_2$$

(11)

in which R = alkyl; and

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 $R_1 = H$ , alkyl, tigloyl, benzoyl, or any other organic ester group;

 $R_2$  = H, or one or more 6-deoxy carbohydrates, or one or more 2,6-dideoxy carbohydrates, or glucose molecules, or combinations thereof;

and in which the broken lines indicate the optional presence of a further bond between C4 - C5, C5 - C6 or C14 - C15.

(12)

in which R = alkyl; and

 $R_1 = H$ , alkyl, tigloyl, benzoyl, any other organic ester group;

 $R_2$  = H, or one or more 6-deoxy carbohydrates, or one or more 2,6-dideoxy carbohydrates, or glucose molecules, or combinations thereof;

and in which the broken lines indicate the optional presence of a further bond between C4 - C5, C5 - C6 or C14 - C15.

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$$OR_1$$
 $OR_3$ 
 $OR_2$ 

(13)

in which R = alkyl; and

R<sub>1</sub> = H, alkyl, tigloyl, benzoyl, any other organic ester group;

 $R_2$  = H, or one or more 6-deoxy carbohydrates, or one or more 2,6-dideoxy carbohydrates, or glucose molecules, or combinations thereof;

and in which the broken lines indicate the optional presence of a further bond between C4 - C5, C5 - C6 or C14 - C15; and

 $R_3 = H$ , alkyl, aryl, acyl, or glucoxy.

in which R = H, alkyl, aryl or any steroid possessing a C14 beta hydroxy group, or a C12 beta hydroxy functionality, or a C17 acyl group, or a C5 - C6 olefin, or combinations thereof.

The invention still further extends to a process for synthetically producing a compound having appetite suppressant activity.

The process uses a steroid as a starting material (or intermediate or precursor), the steroid having the chemical formula

The steroid (15) can be prepared from a compound having the formula (22) by a process which includes the steps of

### (i) treating progesterone having the formula

with the micro-organism Calonectria decora to produce a compound 12 $\beta$ , 15 $\alpha$ - dihydroxy progesterone of the formula

(17)

(ii) treating compound (17) with tosyl chloride and pyridine to produce a compound 12%-hydroxy-15 $\alpha$ -(ptoluene sulfonyl)-progesterone of the formula

(iii)

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treating the compound (18) with collidine at  $150\,^{\circ}$ C to produce a compound  $12\text{B-hydroxy-}\Delta^{14}$ -progesterone of the formula

(19)

(iv) treating the compound (19) with acetyl chloride and acetic anhydride at 120°C, to produce a compound 3,12ß-diacetoxypregna-3,5,14-trien-20-one of the formula

(20)

5 (v) treating the compound (20) with ethylene glycol and a catalytic amount of p-toluene sulphonic acid, to produce a compound 3,12ß-diacetoxy-20,20-ethylenedioxypregna-3,5,14-triene of the formula

(21)

(vi) treating the compound (21) with NaBH<sub>4</sub> to produce a compound 3ß, 12ß-dihydroxy-20,20-ethylenedioxypregna-5,14-diene-12-acetate of the formula

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In a first alternative procedure, a process for the preparation of steroid (15) according to the invention includes the steps of

(a) treating compound (22) with a reducing agent, e.g. LiAlH<sub>4</sub>, to produce a compound 3ß, 12ß-dihydroxy-20,20ethylenedioxypregna-5,14-diene of the formula

(23)

(b) treating compound (23) with N-bromoacetamide (NBA) and a base, e.g. pyridine, to produce a compound 3ß,12ß-dihydroxy-14,15-epoxy-20,20-ethylenedioxypregn-5-ene of the formula

(c) treating compound (24) with a reducing agent, e.g. LiAlH<sub>4</sub>, e.g. with refluxing, to produce a compound 3ß, 12ß, 14ß-trihydroxy-20,20-ethylenedioxypregn-5-ene of the formula

(25)

5 and (d) treating compound (25) with an acid, e.g. acetic acid, and water to produce the steroid intermediate compound 3g, 12g, 14g-trihydroxy-pregn-5-ene (15).

Reaction Scheme A depicts the procedure for the preparation of steroid intermediate (15) from compound (22) according to "the first alternative procedure" of the invention (and includes the preparation of compound (22) from compound (16) for illustrative purposes).

# Reaction Scheme A

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In a second alternative procedure, a process for the preparation of steroid (15) according to the invention includes the steps of

(a) treating compound (22) (3β, 12β-dihydroxy-20,20ethylenedioxypregna-5,14-diene-12-acetate) with ptoluenesulfonyl chloride and a base, e.g. pyridine, to produce a compound 3β, 12β-dihydroxy-20,20ethylenedioxypregna-5,14-diene-3-tosyl-12-acetate of the formula

(b) treating compound (26) with potassium acetate in a solvent, e.g. acetone, to produce a compound 6β, 12βdihydroxy-20,20-ethylenedioxy-3,5α-cyclopregnan-14-ene-12-acetate of the formula

(c) treating the compound (27) with a reducing agent, e.g. LiAlH<sub>4</sub>, and e.g. tetrahydrofuran, to produce a compound  $6\beta$ ,  $12\beta$ -dihydroxy-20,20-ethylenedioxy-3,5 $\alpha$ -cyclopregnan-14-ene of the formula

(28)

(d) treating the compound (28) with N-bromoacetamide, optionally acetic acid, and a base, e.g. pyridine, to produce a compound  $6\beta$ ,  $12\beta$ -dihydroxy-20,20-ethylenedioxy-14,15-epoxy-3,5 $\alpha$ -cyclopregnane of the formula

(29)

10 (e) treating the compound (29) with a reducing agent, e.g. LiAlH<sub>4</sub>, and e.g. tetrahydrofuran, to produce a compound  $6\beta$ ,  $12\beta$ ,  $14\beta$ -trihydroxy-20,20-ethylenedioxy-3,5 $\alpha$ -

#### cyclopregnane of the formula

(30)

and (f) treating compound (30) with an acid, e.g.
hydrochloric acid, and a solvent e.g. acetone, to
produce compound (15).

Reaction Scheme B shows the procedure for the preparation of steroid intermediate (15) from compound (22) according to "the second alternative procedure" of the invention.

## Reaction Scheme B

Compound (1) may be synthesized from a first carbohydrate intermediate in the form of an activated monosaccharide cymarose moiety, which can be prepared from a compound having the formula (36). Compound (36) can be prepared by a process which includes the steps

(i) treating methyl- $\alpha$ -D-glucose having the formula

(31)

with benzaldehyde and zinc chloride to produce a compound methyl-4,6-0-benzylidene- $\alpha$ -D-glucopyranoside of the formula

(32)

(ii) treating the compound (32) with tosyl chloride and pyridine at 0°C, to produce a compound methyl-4,6-0-benzylidene-2-0-tosyl- $\alpha$ -D-glucopyranoside of the formula

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(33)

(iii) treating the compound (33) with NaOMe at 100°C to produce a compound methyl 4,6-0-benzylidene-3-0-methyl- $\alpha$ -D-altropyranoside of the formula

(34)

(iv) treating the compound (34) with N-bromosuccinamide (NBS) to produce a compound methyl 6-bromo-4-0-benzoyl-3-0-methyl-6-deoxy- $\alpha$ -D-altropyranoside of the formula

(35)

and (v) treating the compound (35) with NaBH $_4$  and NiCl $_2$ , to produce a compound methyl 4-0-benzoyl-3-0-methyl-6-deoxy- $\alpha$ -D-altropyranoside of the formula

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(36)

The invention extends to a process for the preparation of a carbohydrate intermediate in the form of an activated monosaccharide cymarose moiety which includes the steps of

(i) treating the compound (36) with PhSSiMe<sub>3</sub>, ZnI<sub>2</sub> and Bu4<sup>+</sup>I<sup>-</sup> to produce a compound 4-0-benzoyl-3-0-methyl-6deoxy-αβ-D-phenylthioaltroside of the formula

(37)

(ii) optionally treating the compound (37) with diethylaminosulphur trifluoride (DAST), e.g. at 0°C, to produce a compound 4-0-benzoyl-3-0-methyl-2phenylthio-2,6-dideoxy-αβ-D-fluorocymaropyranoside having the formula

(38)

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or (iii) optionally, treating the compound (37) with t-butyldimethylsilylchloride and imidazole in a solvent, e.g. pyridine, to produce 4-0-benzoyl-3-0-methyl-2-0-t-butyldimethylsilyl- $\alpha\beta$ -D-phenylthioaltroside having the formula

(39)

in which Z = TBDMS = t-butyldimethylsilyl

and (iv) treating the compound (39) with a base, e.g. sodium methoxide, to produce 3-0-methyl-2-0-t-butyldimethylsilyl- $\alpha\beta$ -D-phenylthicaltroside having the formula

(40)

in which Z = TBDMS = t-butyldimethylsilyl.

Reaction Scheme C shows the procedure for the synthesis of the activated monosaccharide cymarose moiety (40) from compound (36) according to the invention (and includes the preparation of compound (36) from compound (31) for illustrative purposes).

# Reaction Scheme C

Z=t-butyldimethylsilyl

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The synthesis of compound (1) may also involve a second carbohydrate intermediate in the form of an activated monosaccharide thevetose moiety, which can be prepared from a compound having the formula (47). Compound (47) can be prepared by a process which includes the steps of

(i) treating  $\alpha$ -D-glucose having the formula

(41)

with acetone and sulphuric acid to produce a compound 1,2 : 5,6-di-0-isopropylidene- $\alpha$ -D-glucofuranose of the formula

(42)

10 (ii) treating the compound (42) with NaH and MeI to produce a compound 1,2:5,6-Di-0-isopropylidene-3-0-methyl-α-D-glucofuranose of the formula

(43)

(iii) treating the compound (43) with acetic acid to produce a compound 3-0-methyl- $\alpha$ G-D-glucopyranose of the formula

(44)

(iv) treating the compound (44) with methanol and hydrochloric acid to produce a compound methyl 3-0methyl-αβ-D-glucopyranoside having the formula

(45)

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(v) treating the compound (45) with benzaldehyde and zinc chloride to produce a compound methyl 4,6-0-benzylidene-3-0-methyl- $\alpha$ ß-glucopyranoside having the formula

(46)

(vi) treating the compound (46) with N-bromosuccinamide, nickel chloride and sodium borohydride to produce a compound methyl 4-0-benzoyl-3-0-methyl-6-deoxy-αβglucopyranoside having the formula

(47)

The invention extends to a process for the preparation of an activated monosaccharide thevetose moiety which includes the steps of

(i) treating the compound (47) with phenylthiotrimethylsilane and trimethylsilyltrifluoromethanesulphonate to produce a compound 4-0-benzoyl-3-0-methyl-1-phenylthio-6deoxy-αβ-glucopyranoside having the formula

(48)

(ii) treating the compound (48) with pivaloyl chloride and a solvent, e.g. pyridine, to produce a compound 4-0-benzoyl-3-0-methyl-2-0-pivaloyl-1-phenylthio-6-deoxy- $\alpha\beta$ -glucopyranoside having the formula

(49)

and (iii) treating the compound (49) with a brominating agent, e.g. N-bromosuccinimide, and diethylaminosulphur trifluoride to produce a compound 4-0-benzoyl-3-0-methyl-2-0-pivaloyl-1-fluoro-6-deoxy-β-glucopyranoside occurring as stereo-isomers having the formula

Reaction Scheme D shows the procedure for the synthesis of the activated monosaccharide thevetose moiety (50(A) and 50(B)) from compound (48) according to the invention (and includes the preparation of compound (47) from compound (41) for illustrative purposes).

### Reaction Scheme D

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According to a still further aspect of the invention there is provided a process of synthetically producing a compound of the formula (1) and analogues and derivatives thereof which includes the steps of synthesising a suitable steroid intermediate or precursor and coupling the required number of suitable monosaccharides with the steroid intermediate.

The invention also provides a process of coupling a monosaccharide cymarose with the steroid intermediate, which includes the steps of

(i) reacting a cymarose moiety (38) with a steroid intermediate (15), e.g. at -15°C, and in the presence of tin chloride, in a solvent, e.g. ether, to produce a compound 3-0-[4-0-benzoyl-2-phenylthio-ß-D-cymaropyranosyl]-12,14-ß-dihydroxy-pregn-5-ene-20-one of the formula

and (ii) treating the compound (51) with tiglic acid chloride in pyridine and thereafter with a base, e.g. NaOMe, to produce a compound 3-0-[-2-phenylthio-ß-D-cymaropyranosyl]-12ß-tigloyloxy-14-hydroxy-14ß-pregn-5-ene-20-one of the formula

(51)

(52)

The invention extends to a process which includes coupling a monosaccharide cymarose moiety to a monosaccharide thevetose moiety and coupling the resultant disaccharide with the combined steroid product (52) to form compound (1).

The process of coupling the monosaccharide cymarose moiety to the monosaccharide thevetose moiety and coupling the resultant disaccharide to the combined steroid product (52) may include the steps of

(i) coupling a selectively protected cymarose moiety (40) and a selectively protected thevetose moiety (50 A) using tin chloride (SnCl<sub>2</sub>) and silver trifluoromethanesulphonate, e.g. at -15°C, to produce a compound of the formula

in which Z = TBDMS = t-butyldimethylsilyl

(ii) treating compound (53) with tetrabutylammoniumfluoride to produce a compound of the formula

5 (iii) treating compound (54) with diethylaminosulphur trifluoride, e.g. at 0°C, to produce a compound of the formula

(iv) reacting compound (55) with compound (52) to produce
 a compound of the formula

and (v) treating compound (56) in a Raney-Nickel reaction and thereafter with a base, e.g. NaOMe, to produce compound (1) as described above.

Reaction Scheme E shows the procedure for the synthesis of intermediates (52) and (55) and coupling them to form compound (56).

# Reaction Scheme E

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According to the invention, an alternative process is provided which includes coupling cymarose and thevetose moieties to form a trisaccharide and coupling the trisaccharide onto a steroid derivative to form a compound of the formula (1).

The process of forming the trisaccharide and coupling the resultant trisaccharide to a steroid derivative may include the steps of

(i) coupling a selectively protected cymarose moiety (40) and compound (45) using tin (II) chloride, AgOTf, Cp<sub>2</sub>ZrCl<sub>2</sub> to produce a compound of the formula

(57)

in which Z = TBDMS = t-butyldimethylsilyl

(ii) treating compound (57) with tetrabutylammoniumfluoride and diethylaminosulphur trifluoride to produce a trisaccharide compound having the formula

(58)

and (iii) coupling the trisaccharide (58) with a steroid intermediate of the formula

(59)

using tin (II) chloride, AgOTf,  ${\rm Cp_2ZrCl_2}$  to produce compound (1).

5 The steroid intermediate (59) may be produced by treating steroid (15) with tiglic acid chloride.

Reaction Scheme F shows the procedure for the synthesis of the trisaccharide (58) and the synthesis of compound (1) by coupling the trisaccharide (58) with the steroid intermediate (59).

# Reaction Scheme F

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The intermediates (23), (24), (25), (27), (28), (29), (30), (37), (38), (39), (40), (48), (49), (50), (51), (53), (54), (55), (56), (57) and (58) described above are novel compounds and the invention extends to these compounds as such.

Compound (1),  $3-0-[-\beta-D-thevetopyranosyl-(1\rightarrow4)-\beta-D-cymaropyranosyl-(1\rightarrow4)-\beta-D-cymaropyranosyl]-12<math>\beta-0-tigloyloxy-14-hydroxy-14\beta-pregn-5-en-20-one$ , and various analogues and derivatives thereof have been found to have appetite suppressing activity.

The invention extends also to a composition or formulation having appetite suppressant activity, in which the active ingredient is an extract obtained from a plant of the genus *Trichocaulon* or the genus *Hoodia*.

The active ingredient may be a compound of the formula (1), extracted from a plant of the genus Trichocaulon or Hoodia or a derivative thereof. The plant may be of the species Trichocaulon officinale or Trichocaulon piliferum, or the species Hoodia currorii, Hoodia gordonii or Hoodia lugardii.

The invention extends also to a composition or formulation having appetite suppressant activity, in which the active ingredient is a synthetically produced compound of the formula (1) or a derivative or analogue thereof, as hereinbefore set out with reference to compounds (2) to (14).

According to another aspect of the invention there is provided a method of suppressing an appetite by administering to a human or animal a suitable dosage of an appetite suppressant agent comprising an extract of a plant of the genus *Trichocaulon* or *Hoodia*. The extract may be incorporated in a composition or formulation including also pharmaceutically acceptable other ingredients.

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The appetite suppressant agent may be an isolated natural chemical or a synthetic chemical compound of the formula:

(1)

or derivatives or analogues thereof, as set out before.

The appetite suppressant composition or formulation may consist of the appetite suppressant agent admixed with a pharmaceutical excipient, diluent or carrier. Other suitable additives, including a stabilizer and such other ingredients as may be desired may be added.

The invention extends to the use of compound (1) or its derivatives or analogues in the manufacture of a medicament having appetite suppressant activity.

The invention further extends to compound (1), or its derivatives or analogues as set out before, for use as a medicament having appetite suppressant activity.

A method of suppressing an appetite by administering to a human or animal an effective dosage of a composition as described above is also provided.

A method has been described herein for extracting a steroidal glycoside having appetite suppressant activity from plant material obtained from a plant of the

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Trichocaulon or Hoodia genus. The invention thus extends to an extract obtained from plant material of the Trichocaulon or Hoodia genus and containing a substantially pure steroidal glycoside of formula (1).

The invention extends also to a foodstuff or a beverage containing an effective quantity of the steroidal glycoside of the formula (1), or its derivatives or analogues as set out before, to have an appetite suppressant effect when ingested.

Molecular genetic studies have led to a considerable increase in the understanding of the regulation of appetite, satiety and bodyweight. These studies have revealed numerous central regulatory pathways, mediated by a number of neuropeptides. The maintenance of a normal body weight is achieved by an intricate balance between energy intake, food consumption, and energy expenditure. homeostasis is subject to a wide range influences, ultimately controlled by the brain. different signals include such things as sense of smell and taste and gastro-intestinal signals such as distension of the gastro-intestinal tract, chemical signals to gastric mucosa and blood-borne metabolites such as fatty acids and glucose.

Centrally, neuropeptide "Y" (NPY) which is negatively regulated by leptin, has been established as one of the positive regulators of feeding behaviour. Expression of the endogenous antagonist for melanocortin receptors has also been shown to be the basis for obesity in a particular model (the ob/ob mouse). Indeed deficiency at the MC4 melanocortin receptor completely replicates the obesity syndrome. Other mediators which have been shown to have roles in the energy balance include bombesin, galonin and glucagon-like peptide-1.

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Without being bound by theory, the Applicant believes that compound (1) and its analogues as described above act as an agonist of the melanocortin 4 receptor. The effect of this is to regulate NPY but also to increase cholecystokinin. The effect of cholecystokinin amongst other things is to inhibit gastric emptying.

Accordingly, the invention extends to a composition having appetite suppressant activity comprising a melanocortin 4 receptor agonist.

The agonist may be an extract or compound as previously described, in particular the compound of formula (1). The composition may be admixed with a pharmaceutical excipient, diluent or carrier and is optionally prepared in unit dosage form.

The invention still further extends to the use of a melanocortin 4 receptor agonist in the manufacture of a medicament having appetite suppressant activity, to a melanocortin 4 receptor agonist for use as a medicament having appetite suppressant activity, to a method of suppressing an appetite by administering to a human or animal an effective dosage of a composition comprising a melanocortin 4 agonist as described above, and to the use of a melanocortin 4 receptor agonist to suppress the appetite of and/or to combat obesity in a human or animal.

The invention and its efficacy will now be further described, without limitation of the scope of the invention, with reference to the following examples and drawings.

In the drawings,

Figure 1 shows a flow diagram of the general method of extracting a first crude appetite suppressant extract and a purified appetite suppressant extract from plant material of the genus Trichocaulon or Hoodia;

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Figure 2 shows a graphical representation of a bioassay carried out on rats using a partially purified methanol extract of *Trichocaulon piliferum*;

Figures 3 and 4 together show a schematic representation of a preferred embodiment of the process of the invention for producing an extract of plant material of the genus *Trichocaulon* or *Hoodia*; and

Figures 5 and 6 show a graphical representation of the percentage change of body mass of rats for different groups for days -7 to 7 and days 0 to 7 respectively in a repeat dose study using a sap extract and a spray-dried sap extract of plant material of the species *Hoodia gordonii*.

#### EXAMPLE 1

The general method of extracting a first crude appetite suppressant extract and a purified appetite suppressant extract from plant material of the genus *Trichocaulon* or of the genus *Hoodia* is illustrated by way of the flow diagram of Figure 1.

#### EXAMPLE 2

Bioassays carried out on rats using a partially purified methanol extract obtained in the manner illustrated in Example 1, indicated that the extract does in fact exhibit appetite suppressant activity. The appetite suppressant activity of the active extract can be illustrated by way of a typical example of the effect of the methanol extract of *Trichocaulon piliferum* on rats, by way of the graphic representation in Figure 2.

It will be evident from Figure 2 that the test group of rats dosed with the extract on day 5 displayed a substantially diminished food intake over the next two days, while a control group did not disclose a comparable

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reduced food intake. The food intake of the test group returned to normal, and in fact increased, from day 8 onwards.

#### EXAMPLE 3

A preferred embodiment of a process in accordance with the invention for producing an extract having appetite suppressant activity is illustrated schematically by way of example in Figures 3 and 4, which two Figures together illustrate the comprehensive process. However, various other procedures may be used, as will be understood by persons skilled in the art.

Referring to Figure 3, plant material of the genus Trichocaulon or the genus Hoodia is fed into a blender 3, eg a Waring blender, by way of feedline 1, with a solvent in the form of a methylene chloride/methanol solution introduced via feedline 2. The homogenised product is fed via line 4 into a separation stage 5, eg in the form of a filter or centrifuge, and the residual plant material is removed via line 27.

The solvent/extract mixture is fed via line 6 into an evaporation stage 7, where the solvent is removed, for example by means of a rotor evaporator. The dried crude extract is fed via line 8 into a further extraction stage 9 with the addition of a methylene chloride/water solution introduced via feedline 29 for further extraction, and then to a separation stage 13 by way of line 11, where the water fraction is removed via line 31. The dissolved extract fraction is fed via line 15 into a drier stage 17 where the solvent is evaporated, for example by a rotor evaporator.

Referring to Figure 4, the dried extract is fed via line 10 into an extraction stage 12. A methanol/hexane solution is also fed via line 14 into the extraction stage 12 for further purification and extraction of the dried

extract. The extract/methanol/hexane mixture is fed via line 16 into a separation stage 18, the hexane fraction is removed via line 20, and the methanol/extract mixture is then fed via line 22 into a drying stage 24. In the drying stage 24, the solvent is removed, eg by evaporation on a rotor evaporator.

The dried, partially purified active extract is fed via line 26 and with the addition of methanol via line 28 into a solution stage 30, and the dissolved fraction is fed via line 36 to a chromatography column 38.

In the column 38 the methanol soluble fraction is further fractionated, using silica gel and a chloroform/30% methanol solvent, into different fractions schematically indicated as fractions I to V. According to an actual fractionation procedure carried out by the Applicant, the fractionation procedure yielded the following fraction weights : I(3.9 g); II(2.6 g); III(2.1 g); IV(1.1 g) and V(2.0 g). These fractions are individually evaluated by a suitable bioassaying procedure (in a step not shown) and fractions identified as fractions I and displaying marked appetite suppressant activity, are fed by feedlines 40 and 42 into columns 44 and 46 respectively where they are further fractionated and purified by column chromatography, again by using silica gel and a 9:1 chloroform: methanol system.

The sub-fractions II(A) - (C) obtained from column 44 do not, when assayed, display a noteworthy appetite suppressant activity, and may be recycled for further chromatography.

The sub-fractions I(A) - (L) obtained from column 46 are also evaluated (by an assaying step not shown), and the sub-fraction I(C) is found to have marked appetite suppressant activity.

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The sub-fraction I(C) is fed via line 48 into column 50 for a further fractionation and purification, using silica gel and a 9:1 ethyl acetate:hexane eluent. Of the resultant purified fractions, fraction I(C)(ii) is found, after assaying, to possess marked appetite suppressant activity.

The purified product is identified by nuclear magnetic resonance spectroscopy (as indicated in Tables 1 and 2 below), to be compound (1).

Table 1:  $^{1}\text{H}$  (300.13 MHz) n.m.r. data for compound (1) CDC $\ell_{3}$ 

	Compound (1)	
Hydrogen Atom	J(HH)/Hz	δ <sub>H</sub> /p.p.m.
Aglycone-3 6 12 17 18 19 21 3 4 5	- 11.5, 4.1 9.3, 9.3 - - - 7.1, 1.5 7.1, 1.2 1.6, 1.2	3.522 m 5.381 m 4.607 dd 3.157 dd 1.029 s 0.951 s 2.164 s 6.888 qq 1.806 dq 1.853 dq
Cym-1' 2' <sub>aq</sub> 2'ax 3' 4' 5' 6'	9.4, 2.1 13.8, 3.7, 2.1 13.8, 9.4, 2.6 3.7, 2.9, 2.6 9.4, 2.9 6.3, 9.4 6.3	4.816 dd 2.055 ddd 1.552 ddd 3.776 ddd 3.179 dd 3.821 dd 1.279 d <sup>a</sup>
3'-OMe 1'' 2'' 2''aq 3''ax 4'' 5'' 6'' 3''-OMe	9.4, 2.1 13.8, 3.7, 2.1 13.8, 9.4, 2.6 3.7, 2.9, 2.6 9.4, 2.9 6.3, 9.4 6.3	3.408 s <sup>d</sup> 4.730 dd 2.108 ddd 1.601 ddd 3.755 ddd 3.239 dd 3.898 dd 1.243 d <sup>b</sup> 3.392 s <sup>e</sup>
Thev-1''' 2''' 3''' 4''' 5''' 6''' 3'''-OMe	7.7 7.7, 8.0 8.0, 2.9 9.3, 2.9 6.3, 9.3	4.273 d 3.469 dd 3.099 dd 3.179 dd 3.351 dd 1.183 d' <sup>C</sup> 3.622 s

a,b,c in each column may be interchangeable. d,e in each column may be interchangeable,

Refers to the tigloate group atoms

Table 2: Relevant  $^{13}$ C (75.25 MHz) n.m.r. data for Compound (1) in CDC $\ell_3$ 

	Aglycone moiety		Sugar moiety	
Carbon	$\delta_{ m c}$ /p.p.m.	Carbon		$\delta_{ m c}$ /p.p.m.
1	37.04 T	cym-	1'	95.84 D
2	29.44 T		2'	35.57 T
3	77.24 D		3′	77.05 D
4	38.62 T		4′	82.57 D
5	138.95 S		5′	·68.48 D
6	131.90 D		6′	18.14 Q
7	27.30 T 3'-OI		le	57.93 Q
8	35.30 D		1"	99.54 D
9	43.04 D		2′′	35.17 T
. 10	37.22 S		3''	76.99 D
11	26.04 T		4′′	82.52 D
12	75.88 D		5′′	68.30 D
13	53.71 S		6′′	18.36 Q
14	85.69 S	3''-OMe		57.09 Q
15	34.36 T	Thev-	1′′′	104.28 D
16	24.31 T		2′′′	74.62 D
17	57.18 D		3′′′	85.30 D
18	9.85 Q		4′′′	74.62 D
19	19.27 Q		5′′′	71.62 D
20	216.85 S		6′′′	17.75 Q
21	33.01 Q	3′′′-O	Ме	60.60 Q
1*	167.60 S			
2*	128.69 D			
3*	137.66 D			
4*	14.41 Q			
5*	12.08 Q			

Refers to the tigloate group atoms

# Compound (1)

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IR data: 3440 \text{ cm}^{-1} (OH), 2910 cm<sup>-1</sup> (CH), 1700 cm<sup>-1</sup> (C=0) \left[\alpha_D\right]^{20}_{589} = 12,67^{\circ} (C=3, CHCl<sub>3</sub>) m.p. 147^{\circ}\text{C} - 152^{\circ}\text{C}
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Examples 4 to 13 illustrate the synthetic procedures whereby the intermediate compounds and steroid (15) may be prepared according to "the first alternative procedure".

#### EXAMPLE 4

#### 5 <u>12β, 15α-Dihydroxy progesterone</u> (17)

Cultures of <u>Calonectria decora</u> (ATCC 14767) are prepared by the inoculation of a culture medium comprised of sucrose (900 g), K<sub>2</sub>HPO<sub>4</sub> (30 g), Czapek concentrate (300 ml), corn steep liquor (300 ml) and distilled water (30 l) (150 X 500 ml flasks). After 5 days of shaking at 26°C, progesterone (16) (150 g) in a suspension of Tween 80 (0,1% soln., 1,5 l) is added to the flasks. The cultures are incubated for a further 5 days and then worked-up by centrifugation, decantation, extraction of the medium with chloroform, and then evaporation to yield the dihydroxy progesterone (17) (75 g, 45 %).

<sup>1</sup>H NMR (CDCl<sub>3</sub>) : 5,71 (1H, s, H-4); 4,12-4,22 (1H, m, H-15)

4,43 (1H, br, s, OH); 3,46-3,53 (1H, dd, J = 4,6Hz, H-12); 2,16 Hz (3H, s, H-21); 1,18 (3H, s, H-19); 0,74 (3H, s, H-18)

#### EXAMPLE 5

### 12β-Hydroxy-15α-(p-toluene sulfonyl)-progesterone (18)

The dihydroxy progesterone (17) (75 g, 0.22 mol) is dissolved in dry pyridine (300 ml) and cooled to 0°C. p-Toluene sulfonyl chloride (46 g, 0,24 mol) in dry pyridine (200 ml) is added dropwise to the reaction mixture at 0°C. The reaction is stirred overnight at 0°c, and quenched by the addition of  $\rm H_2O$  (500 ml). The water layer is extracted with ethyl acetate (1  $\ell$ ), and the organic extract washed with hydrochloric acid (6M, 3 X 1  $\ell$ ), aqueous saturated sodium bicarbonate (500 ml), aqueous saturated sodium

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chloride (500 ml), and water (500 ml). The organic layer is dried (MgSO<sub>4</sub>), filtered and evaporated to yield p-toluene sulfonated progesterone (18) (98 g, 92 %) as a viscous dark yellow oil.

5 <sup>1</sup>H NMR (CDCl<sub>3</sub>): 7,7 (2H, d, J = 14Hz, H-2,6); (2H, d, J = 8,4Hz, H-3,5);5,67 (1H, s, H-4); 4,86-4,93 (1H, m, H-15); 3,45-3,50 (1H, dd, J = 4,6Hz, 2,44 (3H, s, H-4Me); 2,15 (3H, s, H-21) 1,13 (3H, s, H-19); 0,74 (3H, s, H-18).

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#### EXAMPLE 6

# 12ß-Hydroxy-A<sup>14</sup>-progesterone (19)

A solution of the tosylated progesterone (18) (98 g, 0,19 mol) in 2,4,6-trimethyl collidine (500 ml) is refluxed The reaction mixture is cooled and at 150°C for 3 h. poured into water (500 ml). The water layer is extracted with ethyl acetate (1  $\ell$ ), after which the organic layer is washed with hydrochloric acid (6M, 3  $\times$  1  $\ell$ ), saturated sodium bicarbonate (500 ml), aqueous saturated sodium chloride (500 ml), and water (500 ml). After drying (MgSO<sub>4</sub>) and filtering, the ethyl acetate is evaporated and the crude mixture is purified by silica gel chromatography, eluting with acetone:

25 chloroform (1:10) to afford  $\Delta^{14}$ -progesterone (19) (50 g, 78 %) as a dark red oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>): 5,73 (1H, s, H-4), 5,28 (1H, dd, J = 2,2Hz, H-15), 4,41 (1H, br, s, OH), 3,49-3,52 (1H, dd, J = 4,3Hz, H-12), 2,80-2,84 (1H, dd, J = 9,2Hz, H-17), 2,14 (3H, s, H-12) 21), 1,19 (3H, s, H-19), 0.89 (3H, s, H-18).

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#### EXAMPLE 7

### 3,12ß-Diacetoxypregna-3,5,14-trien-20-one (20)

A solution of  $\Delta^{14}$ -progesterone (19) (50 g, 0,15 mol) in acetyl chloride (1,5  $\ell$ ) and acetic anhydride (750 ml) is refluxed for 2 hours. The reaction mixture is poured into cold ethyl acetate (1  $\ell$ ) and aqueous saturated sodium bicarbonate is added with stirring until the effervescence The ethyl acetate layer is separated from the sodium bicarbonate layer and washed with further portions of aqueous sodium bicarbonate (3 X 700 ml), thereafter with aqueous saturated sodium chloride (700 ml) and finally with The organic layer is dried (MgSO<sub>4</sub>), water (700 ml). filtered and evaporated to afford the 3,12ßdiacetoxypregna-3,5,14-trien-20-one (20) (60 g, 93 %) as an orange oil.

<sup>1</sup>H NMR(CDCl<sub>3</sub>): 5,68 (1H, s, H-4), 5,44 (1H, m, H-6), 5,31 (1H, dd, J = 2,2Hz, H-15), 4,82-4,86 (1H, dd, J = 4,5Hz, H-12), 3,10-3,18 (1H, t, J = 9,5Hz, H-17), 2,18 (3H, s, 3-Ac), 2,11 (3H, s, 12-Ac), 2,08 (3H, s, H-21), 1,02 (3H, s, H-19), 1,01 (3H, s, H-18)

## EXAMPLE 8

# 3,12ß-Diacetoxy-20,20-ethylenedioxypregna-3,5,14-triene (21)

The diacetoxy compound (20) (60 g, 0,14 mol) is dissolved in benzene (1 l) and ethylene glycol (60 ml) and p-toluene sulfonic acid (1 g) are added. (The benzene is previously refluxed with a Dean-Stark trap). The mixture is refluxed with stirring and azeotropic removal of water for 16 hours. Aqueous saturated sodium bicarbonate solution (500 ml) is added to the cooled solution. This is then washed with brine (500 ml), and with water (500 ml), and dried (MgSO<sub>4</sub>). The solvent is evaporated and the crude mixture purified by silica gel column chromatography,

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eluting with ethyl acetate: hexane (2:8) to yield the ethylenedioxypregna-3,5,14-triene (21) (35 g, 53 %).

<sup>1</sup>H NMR (CDCl<sub>3</sub>): 5,68 (1H, s, H-4), 5,45 (1H, m, H-6), 5,31 (1H, dd, J = 2.2Hz, H-15), 4,73-4,85 (1H, dd, J = 4.4Hz, H-12), 3,78-3,98 (4H, m, ethylenedioxy), 2,16 (3H, s, 3-Ac), 2,04 (3H, s, 12-Ac), 1,29 (3H, s, H-21), 1,12 (3H, s, H-19), 1,02 (3H, s, H-18).

#### EXAMPLE 9

3ß-12ß-Dihydroxy-20,20-ethylenedioxypregna-5,14-diene-12-acetate (22)

The dienolacetate (21) (35g, 0,077 mol) is suspended in ethanol (500 ml) and sodium borohydride (2,8g, 0.074 mol) is added at 0°C. The mixture is allowed to warm to room temperature and stirred overnight. Most of the solvent is removed in vacuo and the mixture is diluted with water (500 ml) and extracted with ethyl acetate (500 ml). Work-up followed by chromatography on silica gel with acetone/chloroform (1:10) yields the 3ß-alcohol (22) (25 g, 80 %).

25 EXAMPLE 10

#### 3B, 12B-Dihydroxy-20, 20-ethylenedioxypreqn-5, 14-diene (23)

The 3ß-alcohol (22) (25 g, 60.2 mmol) in dry tetrahydrofuran (300 ml) is added dropwise to a suspension of lithium aluminium hydride (2,7 g, 72,2 mmol) in dry tetrahydrofuran (500 ml). The reaction mixture is stirred

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at room temperature for 24 hours after which water (2,7 ml) is carefully added and stirred for a further 10 min. Sodium hydroxide (15 % soln, 2,7 ml) is then added and the suspension stirred. After 10 min, water (8,1 ml) is added and the suspension stirred for 10 minutes, filtered, dried  $(\text{MgSO}_4)$ , and the solvent evaporated to afford the 3ß, 12ß dihydroxypregna-diene (23) (20 g, 90 %).

1H NMR (CDCl<sub>3</sub>): 5,36 (1H, m, H-6), 5,23 (1H, dd, J =
2,2Hz, H-15), 3,94-4,06 (4H, m, ethylene dioxy), 3,41-3,52

(1H, m, H-3), 3,32-3,36 (1H, dd, J = 4,3Hz, H-12), 1,31
(3H, s, H) 1,01 (3H, s, H-19), 0,96 (3H, s, H-18).

13C NMR (CDCl<sub>3</sub>): 152,4 (c-14), 140,2 (c-5), 121,1 (c-15)
119,7 (c-6), 111,1 (C-20), 79,8 (C-12), 71,6 (C-3),
63,7 and 63,6 (ethylene dioxy), 58,8 (C-17), 19,0 (C-19),
11,9 (C-18).

38. 128-Dihydroxy-14,15-epoxy-20,20-ethylenedioxypregn-5-ene;

3B,12B-Dihydroxy-5,6-epoxy-20,20-ethylenedioxypregn-14-ene

N-Bromoacetamide (211 mg, 1,5 mmol) is added to a stirred solution of the 5,14-diene (23) (500 mg, 1,34 mmol) in acetone (100 ml), acetic acid (2,5 ml), and water (5 ml) at 0°C. After 15 min sodium sulphite (5 % soln, 50 ml) is added to the reaction mixture. The acetone is evaporated, and the aqueous layer extracted with dichloromethane (3 X The organic layer is dried (MgSO<sub>4</sub>), filtered and evaporated. Pyridine (1 ml) is added to the product, and stirred for 0,5 h. Dichloromethane (100 ml) is then added to the reaction mixture, and the dichloromethane is washed with citric acid (5 % soln, 3 X 100 ml), saturated sodium bicarbonate (50 ml), and water (50 ml). The organic layer is dried  $(MgSO_4)$ , filtered and evaporated to give the mixture of 14,15- and 5,6-epoxides (360 mg, 69%) as a white The mixture of epoxides could not be separated by silica gel column chromatography.

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#### EXAMPLE 11

3ß, 12ß-Dihydroxy-14,15-epoxy-20,20-ethylenedioxypregn-5-ene (24)

The mixture of 14,15- and 5,6- epoxides (14,4 g, 37,0 mmol) in dry tetrahydrofuran (200 ml) is added to a suspension of lithium aluminium hydride (1,69 g, 44,4 mmol) in dry tetrahydrofuran (300 ml). The reaction mixture is stirred at room temperature for 24 hours, after which it is worked up as described earlier by the addition of water (1,69 ml), and sodium hydroxide (15 % soln, 1,69 ml). After filtration and evaporation of the solvent, the crude product is purified by silica gel column chromatography using methanol/chloroform (1:9) as solvent to give the unreacted 14,15 epoxy- 20,20-ethylenedioxypregn-5-ene (24) (300 mg, 2,1 %).

<sup>1</sup>H NMR (CDCl<sub>3</sub>): 5,31 (1H, m, H-6), 3,82-3,98 (4H, m, ethylene dioxy), 3,43-3,52 (1H, m, H-3), 3,41 (1H, s, H-15), 3,31-3,35 (1H, dd, J=4,3 Hz, H-12), 1,29 (3H, s, H-21), 1,17 (3H, s, H-19), 1,02 (3H, s, H-18).

 $^{13}$ C NMR (CDCl<sub>3</sub>): 139,8 (C-5), 120,8 (C-6), 112,1 (C-20), 77,2 (C-12), 75,4 (C-14), 61,0 (C-15), 22,3 (C-21), 19,2 (C-19), 9,5 (C-18).

# EXAMPLE 12

### 38, 128, 146-Trihydroxy-20, 20-ethylenedioxypregn-5-ene (25)

The 14,15-epoxide (24) (300 mg, 0,77 mmol) in dry tetrahydrofuran (10 ml) is added to a suspension of lithium aluminium hydride (300 mg, 7,89 mmol) in tetrahydrofuran and the reaction refluxed for 48 h. After the addition of water (0,3 ml), sodium hydroxide (15 % soln, 0,3 ml) and filtration as described earlier, the mixture is purified by silica gel column chromatography using methanol: chloroform

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(1:9) as solvent to give the trihydroxy pregnene (25) (250 mg, 83 %).

<sup>1</sup>H NMR (CDCl<sub>3</sub>):5,38 (1H, m, H-6), 3,98 (4H, m, ethylene dioxy),

3,43-3,53 (1H, m, H-3), 3,25-3,32 (1H, dd, J = 4,1Hz, H-12), 1,32 (3H, s, H-21), 1,01 (3H, s, H-19), 0,98 (3H, s, H-18)

13C NMR CDCl<sub>3</sub>): 139,1 (C-5), 122,1 (C-6), 112,2 (C-20), 85,1 (C-14), 75,1 (C-12), 71,6 (C-3), 23,4 (C-21), 19,4 (C-19), 8,9 (C-18)

#### EXAMPLE 13

#### 38, 128, 148-Trihydroxy-pregn-5-ene (15)

The ethylenedioxypregnene (25) (250 mg, 0,64 mmol) is dissolved in acetic acid (13,4 ml) and water which after freeze drying affords the trihydroxy steroid (15) (200 mg, 89 %), m.p.: 228°-235°C (lit 225°-235°C), M+ 348,  $[\alpha_D]^{20}$  + 35° (lit  $[\alpha_D]^{20}$  + 29°).

13C NMR (CDCl<sub>3</sub>): 217,7 (C-20), 138,9 (C-5), 122,2 (C-6), 85,5 (C-14), 73,6 (C-12), 71,6 (C-3), 57,0 (C-17), 55,1 (C-13), 43,6 (C-9), 42,1 (C-4), 37,3 (C-1), 36,8 (C-10), 35,9 (C-8), 34,5 (C-15), 32,9 (C-21), 31,5 (C-16), 30,1 (C-2), 27,4 (C-7), 24,4 (C-11), 19,4 (C-19),

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1.5

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# 8,3 (C-18).

Examples 14 to 19 illustrate the synthetic procedures whereby the intermediate compounds and steroid (15) may be prepared according to "the second alternative procedure".

5 <u>EXAMPLE 14</u>

20,20-Ethylenedioxy-3β-toluene-p-sulphonyloxy-pregn-5,14-(26) diene-12β-ol acetate Α solution toluenesulphonyl chloride (650 mg, 3.4 mmol) in pyridine (10 ml) was added dropwise to a mixture of the 20,20-Ethylenedioxypregna-5,14-diene-3 $\beta$ ,12 $\beta$ -diol 12-acetate (22) (1.3 g, 3.1 mmol) in pyridine (15 ml) at 0°C. The reaction mixture was left stirring at room temperature for 24 hours after which water was added to the reaction mixture. solution was extracted with ethyl acetate (2x50 ml), the ethyl acetate layer was washed citric acid (5x50 ml), saturated sodium bicarbonate solution (100 ml), saturated sodium chloride solution (100 ml) and water (100 ml). ethyl acetate was dried (MgSO<sub>4</sub>), filtered, and evaporated and purified by flash column chromatography using hexaneethyl acetate (8:2 v/v) as the eluant to give the  $\beta$ -O-tosyl steroid (26), (1.5 g, 84%), as a yellow oil, (Found M 570.271, C<sub>32</sub>H<sub>42</sub>O<sub>7</sub>S requires: M 570.273).

 $\delta_{\rm H}$  1.021 (3H, s, 19-H), 1.131 (3H, s, 18-H), 1.282 (3H, s, 21-H), 2.021 (acetate0CH<sub>3</sub>), 2.431 (3H, s, Ar-CH<sub>3</sub>), 3.883 (4H, m, OCH<sub>2</sub>CH<sub>2</sub>O), 4.750 (1H, dd, <sup>3</sup> J 10.8 Hz, 5.2 Hz, 12-H), 4.890 (1H, m, 30H), 5.281 (1H, dd, <sup>3</sup> J 4.2 Hz, 2.1 Hz, 15-H), 5.388 (1H, m, 6-H), 7.341 (2H, d, <sup>3</sup> J 8.2 Hz, ArH), 7.746 (2H, d, <sup>3</sup> J 8.2 Hz, ArH).

 $\delta_{\rm c}$  13.493Q (C-18), 19.002Q (C-19), 21.612Q (Armethyl)\*, 21.671Q (C-21)\*, 24.175Q (acetate methyl), 63.401T (ethylenedioxy), 63.498T (ethylenedioxy), 71.531S (C-13), 80.912D (C-12), 82.531D (C-3), 111.363S (C-20), 120.881D (C-15), 121.461D (C-6), 123.715-

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133.917 (Aromatic), 139,903S (C-14), 151,722S (C-5), 170.819S (ester carbonyl).

\* may be interchanged

#### EXAMPLE 15

20,20-Ethylenedioxy- $3\alpha$ ,5-cyclo- $5\alpha$ --pregn-14-ene- $6\beta$ , $12\beta$ -5 diol-12-acetate (27) - A solution of 3ß-toluene-psulphonyloxy-pregn-5,14-diene (26) (1.2 g, 2.1 mmol) and potassium acetate (2.2 g, 22.4 mmol) in water (250 ml) and acetone (500 ml) was refluxed at 60 °C for 16 hours. acetone was evaporated and the water was extracted with 10 ethyl acetate (200 ml). The ethyl acetate was dried (MgSO<sub>4</sub>), filtered, and evaporated. Flash chromatographic separation of the mixture using chloroform-acetone (9:1 v/v) as the eluant gave the  $3\alpha,5$ -cyclo derivative (27), (530 mg, 61%) as a yellow oil, (Found M 416.262,  $C_{25}H_{36}O_{5}$ 15 requires : M 416.263).

 $\delta_{\rm H}$  0.288 (1H, dd,  $^3$  J 8.1 Hz, 4.9 Hz, 4-H<sub>a</sub>), 0.477 (1H, dd,  $^3$  J 4.4 Hz, 4.4 Hz, 4-H<sub>b</sub>), 1.025 (3H, s, 19-H), 1.121 (3H, s, 18-H), 1.256 (3H, s, 21-H), 1.989 (3H, s, acetate-CH<sub>3</sub>), 3.302 (1H, dd,  $^3$  J 2.8 Hz 2.8 Hz, 6-H), 3.784-3.947 (4H, m, OCH<sub>2</sub>CH<sub>2</sub>O), 4.721 (1H, dd,  $^3$  J 8.5 Hz, 5.6 Hz, 12-H), 5.232 (1H, dd,  $^3$  J 3.9 Hz, 1.9 Hz, 15-H).

 $\delta_c$  11.678T(C-4), 12.298Q(C-18), 19.971Q (C-19), 23.623Q(C-21), 24.153Q (acetate methyl), 63.700T (ethylenedioxy), 63.788T (ethylenedioxy), 73.591D (C-6), 80.551D (C-12), 111.126S (C-20), 118.778D (C-15), 152.959S (C-14), 170.991S (ester carbonyl).

### EXAMPLE 16

20,20-Ethylenedioxy-3α,5-cyclo-5α-pregn-14-ene-6β,12β-diol
(28) - A solution of the 3α,5-cyclo derivative (27), (500
mg, 1.2 mmol) in tetrahydrofuran (20 ml) was added dropwise

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to a suspension of lithium aluminium hydride (50 mg, 1.3 mmol) in tetrahydrofuran (10 ml). The reaction mixture was stirred for 4 hours and quenched by the addition of water (50  $\mu$ l). After 30 minutes, sodium hydroxide was added (15% solution, 50  $\mu$ l) and stirring continued for a further 30 Water (150  $\mu$ l was added and the reaction mixture filtered. The tetrahydrofuran was dried filtered and evaporated and flash chromatographic purification using chloroform-acetone (8:2 v/v) as the eluant to give the diol (28), (370 mg, 83%) as an oil, (Found M 374.250, C<sub>23</sub>H<sub>34</sub>O<sub>4</sub> requires: M 374.252)

 $\delta_{\rm H}$  0.298 (1H, dd,  $^3$  J 8.1 Hz, 4.9 Hz, 4-H<sub>2</sub>), 0.510 (1H, dd,  $^3$  J 4.4 Hz, 4.4 Hz,4-H<sub>b</sub>), 0.985 (3H, s, 19-H), 1.055 (3H, s, 18-H), 1.325 (3H, s, 21-H), 3.318 (1H, dd,  $^3$  J 3.0 Hz, 3.0 Hz, 6-H),), 3.363 (1H, dd,  $^3$  J 11.4 Hz, 4.2 Hz, 12-H), 4.019 (4H, m, OCH<sub>2</sub>Ch<sub>2</sub>O) 4.622 (1H, s, OH), 5.255 (1H, dd,  $^3$  J 3.9 Hz, 1.9 Hz, 15-H).

 $\delta_{\text{c}}$  11.681T(C-4), 12.243Q(C-18), 19.844Q (C-19), 23.604Q(C-21), 63.620T (ethylenedioxy), 63.733T (ethylenedioxy), 73.569D (C-6), 77.478D (C-12), 111.125S (C-20), 118.702D (C-15), 152.912S (C-14).

#### EXAMPLE 17

20,20-Ethylenedioxy-14,15 $\beta$ -epoxy-3 $\alpha$ ,5-cyclo-5 $\alpha$ ,14 $\beta$ -pregnane-6 $\beta$ ,12 $\beta$ -diol (29) - N-bromoacetamide (150 mg, 1.1 mmol) was added to a solution of the 20,20-ethylenedioxy-3 $\alpha$ ,5-cyclo-5 $\alpha$ -pregn-14-ene-6 $\beta$ ,12 $\beta$ -diol (28) (340 mg, 0.91 mmol) in acetone (20 ml), water (0.25 ml) and acetic acid (0.25 ml) at 0°C. After 15 min., sodium sulphite (5% solution, 20 ml) was added to the reaction mixture. The acetone was evaporated under reduced pressure and the remaining solution was extracted with dichloromethane (3x30 ml). The dichloromethane layer was dried (MgSO<sub>4</sub>), filtered and evaporated to a concentrated volume (50 ml). Pyridine (0.5 ml) was added to the mixture and stirred for a further

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1 hour after which the dichloromethane layer was washed with a citric acid solution (5%, 3x30 ml), saturated sodium bicarbonate solution (30 ml) and water (30 ml). The dichloromethane layer was dried  $(MgSO)_4$ , filtered and evaporated and purified by flash column chromatography using chloroform-methanol (9.5:0.5 v/v) as the eluant to give the epoxide (29) (180 mg, 51% as a foam, (Found M <math>390.245,  $C_{23}H_{34}O_2$  requires: M 390.247).

 $\delta_{\rm H}$  0.287 (1H, dd,  $^3$  J 8.1 Hz, 4.9 Hz, 4-H<sub>a</sub>), 0.501 (1H, dd,  $^3$  J 4.4 Hz, 4.4 Hz,4-H<sub>b</sub>), 0.978 (3H, s, 19-H), 1.048 (3H, s, 18-H), 1.321 (3H, s, 21-H), 3.318 (1H, dd,  $^3$  J 3.1 Hz, 3.1 Hz, 6-H), ), 3.355 (1H, dd,  $^3$  J 11.2 Hz, 4.1 Hz, 12-H), 3.491 (1H, s, 15-H), 4.001 (4H, m, OCH<sub>2</sub>Ch<sub>2</sub>O), 4.901 (1H, s, OH).

 $\delta_{\rm C}$  11.668T(C-4), 11.973Q(C-18), 19.515Q (C-19), 23.519Q(C-21), 59.910D (C-15), 63.601T (ethylenedioxy), 63.713T (ethylenedioxy), 72.501S (C-14), 73.571D (C-6), 77.471D (C-12), 111.085S (C-20).

#### EXAMPLE 18

20,20-Ethylenedioxy-6β,12β,14-trihydroxy-3α,5-cyclo-5α,14βpregnane (30) - A solution of the epoxide (29) (170 mg, 0.44 mmol) in tetrahydrofuran (10 ml) was added to a suspension of lithium aluminium hydride (20 mg, 0.53 mmol) in tetrahydrofuran (5 ml). The reaction mixture was refluxed for 2 hours after which water (20  $\mu$ l) was added and stirring continued for 05 hour. Sodium hydroxide solution (15%,  $20\mu$ l) was added and stirring continued for a further 0.5 hour. A further quantity of water was added (60  $\mu$ l) and the suspension was stirred for 1 hour. After filtration, the suspension was dried (MgSO<sub>4</sub>) filtered, and the tetrahydrofuran was evaporated. Flash chromatographic separation of the resulting mixture eluting chloroform-methanol (9:1 v/v) gave the required triol (30),

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(90 mg, 53%) as a clear oil, (Found M 392.261,  $C_{23}H_{38}O_5$  requires: M 392.263).

 $\delta_{\rm H}$  0.287 (1H, dd,  $^3$  J 8.1 Hz, 4.9 Hz, 4-H<sub>2</sub>), 0.510 (1H, dd,  $^3$  J 4.4 Hz, 4.4 Hz,4-H<sub>b</sub>), 0.971 (3H, s, 19-H), 1.042 (3H, s, 18-H), 1.319 (3H, s, 21-H), 3.321 (1H, dd,  $^3$  J 3.0 Hz, 3.0 Hz, 6-H), 3.321 (1H, dd,  $^3$  J 11.1 Hz, 3.9 Hz, 12-H), 3.561 (1H, s, OH), 4.084 (4h, m, OCH<sub>2</sub>Ch<sub>2</sub>O) 4.671 (1H, s, OH).

 $\delta_c$  11.668T(C-4), 11.971Q(C-18), 19.511Q (C-19), 23.520Q (C-21), 63.612T (ethylenedioxy), 63.711T (ethylenedioxy), 73.483D (C-6), 76.051D (C-12), 84.307S (C-14), 111.099S (C-20).

#### EXAMPLE 19

3B, 12B, 14-Trihydroxy-14B-pregn-5-en-20-one (15) - A mixture of the triol (30) (80 mg, 0.20 mmol) in acetone (20 ml) and hydrochloric acid (1M, 10 ml) was refluxed at 60°C for 2 The reaction mixture was cooled and saturated sodium bicarbonate solution (20 ml) was added. The acetone was evaporated and the aqueous layer extracted with chloroform (3 x 20 ml), the chloroform layer was dried (MgSO<sub>4</sub>), filtered and evaporated to give the epimeric trihydroxy steroids (15a, 15b) (42 mg, 61%). Separation of the epimeric mixture (15a, 15b) (15 mg) was achieved by chromatographic separation using chloroform methanol (9:1 v/v) as the eluant to give the pure 176epimer (15a), (10 mg), m.p. 224-229°C (acetone), (lit. 226-223°), (Found M 348.234, C; 72.32, H 9.21% C<sub>21</sub>H<sub>32</sub>O<sub>4</sub> H 9.26%, M 348.236), and the  $17\alpha$ requires: C, 72.38; epimer (15B) (3 mg), m.p. 183-191°C (acetone), (lit 184-196°).

38,128,14-Trihydroxy-148-pregn-5-en-20-one (15a):  $\delta_{\rm H}$  0.963 (1H, s, 19-H), 1.192 (3H, s, 18-H), 2.236 (3H, s 21-H), 3.325 (1H, dd,  $^3$  J 11.2 Hz, 3.9 Hz, 12-H),

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3.464 (1H, s, OH), 3.5140 (1H, m, 3-H), 3.598 (1H, dd,  $^3$  J 9.6 Hz, 9.6 Hz, 17-H), 4.255 (1H, s, OH), 5.383 (1H, m, 5-H).

δ<sub>c</sub> 8.275Q (C-18), 19.414Q (C-19), 24.400T (C-11) 24.581T (C-16), 27.443T (C-7), 30.062T (C-2), 32.972Q (C-21), 34.543T (C-15), 35.864D (C-8), 36.975S (C-10), 37.337T (C-1), 42.144T (C-4), 43.565D (C-9), 55.101S (C-13), 57.038D (C-17), 71.597D (C-3), 73.558D (C-12), 85.566S (C-14), 122.223D (C-6), 138.932S (C-5), 217.011S (C-20).

3B,12B,14-Trihydroxy-14B-pregn-5-en-20-one (15b):  $\delta_{\rm H}$  0.996 (1H, s, 19-H), 1.144 (3H, s, 18-H), 2.221 (3H, s 21-H), 3.339 (1H, dd,  $^3$  J 9.4 Hz, 9.4 Hz, 17-H), 3.492 (1H, m, 3-H), 3.629 (1H, dd,  $^3$  J 11.1 Hz, 3.9 Hz, 12-H), 3.712 (1H, s, OH), 4.325 (1H, s, OH), 5.383 (1H, m, 5-H).

Examples 20 to 28 illustrate the procedures whereby the intermediate compounds may be prepared to form the first monosaccharide (40).

20 <u>EXAMPLE 20</u>

#### Methyl-4,6-0-benzylidene-α-D-glucopyranoside (32)

A mixture of methyl- $\alpha$ -D-glucopyranoside (30 g, 0,15 mol), benzaldehyde (70 ml) and zinc chloride (20 g) is stirred at room temperature for 24 hours. The reaction product is poured into ice water and stirring continued for 15 min. The white precipitate is filtered and washed with diethyl ether. The solid material is stirred with a solution of sodium metabisulphite (10 % soln), for 15 min, filtered and washed with water. The solid material is crystallized from chloroform and ether to yield the benzylidene product (32) (31 g, 72 %).

#### EXAMPLE 21

# Methyl-4,6-0-benzylidene-2-0-tosyl-α-D-glucopyranoside(33)

p-Toluene sulfonyl chloride (25 g, 1,2 eq) in pyridine (100 ml) is added dropwise to a solution of the benzylidene glucose (32) (31 g, 0.12 mol) in pyridine (100 ml) at 0°C. The reaction is stirred at room temperature for 48 hours. Ice is added to the reaction mixture. The resulting white solid material is washed with water and recrystallized from hot ethanol to yield the tosylated glucose (33) (28 g, 60 왕).

#### EXAMPLE 22

Methyl-4,6-0-benzylidene-3-0-methyl-α-D-altropyranoside (34)

The tosylate (33) (28 g, 64 mmol) in a solution of sodium (7 g) in methanol (150 ml) is heated at 110°C for 48 hour in an autoclave. The reaction vessel is cooled and solid carbon dioxide is added to the reaction mixture. After filtration, the methanol is evaporated and the solid material is then taken up in water. The aqueous layer is extracted with chloroform (X 3). The chloroform is dried (MgSO<sub>4</sub>), filtered and evaporated. The crude mixture is purified by silica gel column chromatography eluting with chloroform : acetone (9:1) to yield the altroside (34) (10 g, 52 %).

25 EXAMPLE 23

> $\underline{\mathsf{Methyl-6-bromo-4-0-benzoyl-3-0-methyl-6-deoxy-\alpha-D-}}$ altropyranoside (35)

> The benzylidene altroside (34) (10 g, 33 mmol) is added to a solution of N-bromosuccinimide (7.6 g) and barium carbonate (20 g) in carbon tetrachloride and the reaction

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mixture is refluxed at 75°C for 3 hours. The reaction mixture is filtered and the carbon tetrachloride layer is washed with water. The organic layer is dried (MgSO4), filtered and evaporated to yield 6-bromo-altroside (35), (9 g, 69 %).

#### EXAMPLE 24

Methyl-4-0-benzoyl-3-0-methyl-6-deoxy-α-D-altropyranoside (36)

Sodium borohydride (18 g) in water (30 ml) is added dropwise to a solution of the bromoaltroside (35) (9 g, 23 mmol) and nickel chloride (18 g) in ethanol (300 ml) at 0°C. The reaction mixture is refluxed at 75°C for 1 hour and then it is filtered. The ethanol is evaporated and the remaining aqueous layer is extracted with chloroform (X 3). The chloroform is dried (MgSO<sub>4</sub>), filtered and evaporated, to yield the 6-deoxy-altroside (36) (5 g, 72 %).

#### EXAMPLE 25

# $4 - 0 - Benzoyl - 3 - 0 - methyl - 6 - deoxy - <math>\alpha B - D$ - phenylthioaltropyranoside (37)

Phenylthiotrimethylsilane ml) (5 and trimethylsilyltrifluoromethane sulphonate (2 ml) are added at 0°C to a solution of the 6-deoxy-altroside (36) (5 g, 17 mmol) in dichloromethane (200 ml). The reaction mixture is stirred at room temperature for 6 hours. Saturated sodium bicarbonate is added to the reaction mixture. The layer is dried (MgSO<sub>4</sub>), filtered and dichloromethane evaporated. The crude mixture is purified by silica gel column chromatography eluting with chloroform : acetone (9:1) to yield the αβ-phenylthioaltroside (37) (4 g, 63 %).

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### EXAMPLE 26

# 4-0-Benzoyl-3-0-methyl-2-phenylthio-2.6-dideoxy-αβ-D-fluorocymaropyranoside (38)

Diethylaminosulphurtrifluoride (0,65 g) is added rapidly to a solution of the  $\alpha$ S-phenylthioaltroside (37) (0,5 g, 1,33 mmol) in dichloromethane at 0°C. The reaction is stirred for 0,5 h at 0°C and then saturated sodium bicarbonate is added. The dichloromethane is separated from the aqueous layer, dried (MgSO<sub>4</sub>), filtered and evaporated to yield the  $\alpha$ S-fluorocymarose (38) (450 mg, 90%).

#### EXAMPLE 27

# $4-0-Benzoyl-3-0-methyl-2-0-t-butyldimethylsilyl-<math>\alpha\beta-D-$ phenylthio-altroside (39)

The 6-deoxy altroside (37) (5 g) is silylated using t-butyldimethylsilylchloride (3 g) and imidazole (3 g) in pyridine (50 ml). The reaction is worked-up by extracting with ethyl acetate, washing the ethyl acetate with hydrochloric acid (6 N), then with sodium bicarbonate, and finally with water. The ethyl acetate layer is dried (MgSO<sub>4</sub>), filtered and evaporated to yield the silylated benzoyl phenylthioaltroside (39) (80 %).

#### EXAMPLE 28

# $3-0-methyl-2-0-t-butyldimethylsilyl-\alpha\beta-D-phenylthioaltroside (40)$

The silylated benzoyl phenylthicaltroside (39) (6 g) is treated with sodium methoxide (100 ml) for 4 hours. The methanol is evaporated and water is added to the reaction. The water layer is acidified (pH 5, ACOH) and extracted with ethyl acetate. The ethyl acetate is washed with

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water, dried  $(MgSO_4)$ , filtered and evaporated to yield silylated methyl phenylthicaltroside (40) (75%).

Examples 29 to 37 illustrate the procedures synthetic whereby the intermediate compounds may be prepared to form the second monosaccharide (50).

#### EXAMPLE 29

# 1.2 : 5.6-Di-O-isopropylidene-a-D-glucofuranose (42)

Sulfuric acid (40 ml) is added dropwise to a solution of  $\alpha$ -D-glucose (41) (50 g, 0,28 mol) in acetone (1  $\ell$ ) at 0°C. The reaction mixture is stirred for 24 h and then it is neutralized using sodium hydroxide (6 M). The acetone is evaporated and the aqueous layer is extracted with chloroform (X2). The chloroform is dried (MgSO<sub>4</sub>) filtered and evaporated. Crystallization from cyclohexane yielded the di-isopropylidene glucose (42) (41 g, 57 %).

### EXAMPLE 30

# 1.2: 5.6-Di-0-isopropylidene-3-0-methyl- $\alpha$ -D-glucofuranose (43)

The  $\alpha$ -D-glucofuranose (42) (41 g, 0,16 mol) in tetrahydrofuran (300 ml) is added dropwise to a suspension of sodium hydride (5 g) in tetrahydrofuran (200 ml). After 0,5 h, methyl iodide (25 g) in tetrahydrofuran (100 ml) is added dropwise to the reaction mixture which is then stirred for 24 h. Water is added to the reaction mixture which is then extracted with ether (X 3). The ether layer is dried (MgSO<sub>4</sub>), filtered and evaporated to yield the methyl protected glucose (43) (38 g, 83 %).

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#### EXAMPLE 31

#### 3-0-Methyl-\alpha\beta-D-glucopyranoside (44)

The methyl diisopropylidene compound (43) (38 g, 0,14 mol) is dissolved in acetic acid (50 %, 700 ml) and the solution refluxed for 18 h. After cooling the acetic acid is evaporated. The crude product is purified by column chromatography eluting with chloroform : methanol : acetone : water (70 : 27 : 2 : 1) to yield 3-0-methyl- $\alpha$ ß-glucopyranoside (44) (13 g, 50 %).

10 EXAMPLE 32

#### Methyl 3-0-methyl- $\alpha\beta$ -D-glucopyranoside (45)

The 3-0-methyl- $\alpha\beta$ -glucopyranoside (44) (10 g) is dissolved in methanol (50 ml) and HCl (conc.)(1 ml) and refluxed overnight. Solid NaHCO<sub>3</sub> is added and the reaction is filtered. The methanol is evaporated to give 1,3-di-0-methyl- $\alpha\beta$ -D-glucopyranoside (45), (95%).

#### EXAMPLE 33

#### Methyl 4, 6-0-benzylidene-3-0-methyl- $\alpha\beta$ -glucopyranoside (46)

The glucopyranoside (45) (8 g) is stirred at room temperature in a solution of benzalaldehyde (20 ml) and zinc chloride (5 g). After 24 hours, ice is added and the aqueous layer is extracted with chloroform. The chloroform layer is dried (MgSO<sub>4</sub>), filtered and evaporated. The benzalaldehyde is removed by vacuum distillation and the product is purified by silica gel column chromatography eluting with acetone:chloroform (0,5:9,5), to yield benzylidene- $\alpha\beta$ -glucopyranoside (46) (60%).

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#### EXAMPLE 34

# Methyl 4-0-benzoyl-0-methyl-6-deoxy-αβ-qlucopyranoside (47)

The benzylidene compound (46) (5 g) is refluxed at 80°C in a mixture of N-bromosuccinimide (3,7 g) and barium carbonate (4 g) in carbon tetrachloride. After 4 hours, the reaction is filtered and the carbon tetrachloride is washed with water, dried (MgSO<sub>4</sub>), filtered and evaporated to give the bromo compound (70%).

The bromo compound (4,3 g) is dissolved in a solution of ethanol (300 ml) and nickel chloride (8,6 g) at 0°C. To this solution, sodium borohydride (8,6 g) in water (50 ml) is added dropwise over a period of 15 minutes. The reaction mixture is refluxed at 100°C for 45 minutes, cooled, filtered and evaporated. Chloroform is added, and the chloroform layer is washed with water, dried (MgSO<sub>4</sub>), filtered and evaporated to give the 6-deoxy sugar (47) (70%).

#### EXAMPLE 35

# 4-0-Benzoyl-3-0-methyl-1-phenylthio-6-deoxy-αβglucopyranoside (48)

The 6-deoxy glucopyranoside (47) (3 g) is dissolved in dichloromethane (50 ml). To this solution. phenylthiotrimethylsilane (2 q) trimethylsilyltrifluoromethanesulphonate (0,2 ml) are The solution is stirred at room temperature added. overnight, after which saturated sodium bicarbonate is added. The dichloromethane layer is dried (MgSO4), filtered The product is purified by silica gel and evaporated. column chromatography eluting with ethyl acetate:hexane (2:8), to give the compound (48) (60%).

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#### EXAMPLE 36

4-0-Benzoyl-3-0-methyl-2-0-pivaloyl-1-phenylthio-6-deoxy- $\alpha\beta$ -glucopyranoside (49)

To a solution of the glucopyranoside (48) (2 g) in pyridine (20 ml), pivaloyl chloride (2 ml) is added. The solution is stirred at room temperature overnight after which water is added. The aqueous layer is extracted with ethyl acetate, and the organic layer is washed with HCl (6 N). The organic layer is dried (MgSO<sub>4</sub>), filtered and evaporated to give the pivaloyl ester (49) (80%).

#### EXAMPLE 37

 $4-0-Benzoyl-3-0-methyl-2-0-pivaloyl-1-fluoro-6-deoxy-\beta-glucopyranoside$  (50)

N-Bromosuccinimide (1,2 g) and diethylaminosulphur trifluoride (1,2 g) are added to a solution of the pivaloyl ester (49) (2 g) in dichloromethane (100 ml) at 0°C. After 1 hour, saturated sodium bicarbonate is added. The dichloromethane layer is dried (MgSO<sub>4</sub>), filtered and evaporated. The  $\beta$ -fluoropyranoside (50) is purified by silica gel column chromatography eluting with ethyl acetate:hexane (2:8), (yield 45%).

Example 38 illustrates the synthetic procedure whereby the compound 3-0- [4-0-benzoyl-2-phenylthio-ß-D-cymaropyranosyl]-12,14ß-dihydroxy-pregnan-5-ene-20-one(51) may be prepared.

#### EXAMPLE 38

3-0-[4-0-benzoyl-2-phenylthio-ß-D-cymaropyranosyl]-12,14ß-dihydroxy-pregn-5-en-20-one (51)

Tin chloride (190 mg, 1 mmol) is added to a solution of 3,12,14 ß-trihydroxy pregnan-5-ene-20-one (15) (100 mg,

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0,28 mmol) and the fluorocymaropyranoside (38) (210 mg, 0,56 mmol), in dry diethyl ether and 4Å molecular sieves at -15°C. The reaction mixture is maintained at -15°C for 3 days. Saturated sodium bicarbonate is added to the reaction mixture. The ether layer is dried (MgSO<sub>4</sub>), filtered and evaporated. The product is purified by silica gel column chromatography eluting with chloroform : methanol (9, 5:0,5) to yield the glycoside (51) (30 mg, 15%).

Examples 39 to 41 illustrate the synthetic procedures whereby the cymarose and thevetose moieties may be coupled.

#### EXAMPLE 39

# Thevetose-cymarose dissaccharide (53)

A solution of thevetose (50 A) (1,5 g), cymarose (40) (1,3 g), and molecular sieves 4Å in dichloromethane is stirred at room temperature for 1 hour. The reaction mixture is cooled to -15°C, and tin (II) chloride (0,8 g) and silver trifluoromethanesulphonate (1,1 g) are added. The mixture is stirred at -15°C for 16 hours, after which triethylamine (0,5

ml) is added. The reaction product is filtered and the dichloromethane is evaporated. The dissaccharide (53) is purified by silica gel column chromatography eluting with ethyl acetate:hexane (2:8), yield 15%.

### 25 EXAMPLE 40

# Thevetose-cymarose dissaccharide (54)

To a solution of the dissaccharide (53) (200 mg) in tetrahydrofuran (20 ml), tetrabutylammonium fluoride (0,4 ml) is added. The mixture is stirred at room temperature for 1 hour, after which saturated sodium bicarbonate is

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added. The reaction mixture is extracted with ethyl acetate and the ethyl acetate layer is dried (MgSO<sub>4</sub>), filtered and evaporated. The dissaccharide (54) is purified by silica gel column chromatography (acetone:chloroform, 0,5:9,5) yield 60%.

#### EXAMPLE 41

# Thevetose-cymarose dissaccharide (55)

To a solution of the dissaccharide (54) (80 mg) in dichloromethane (10 ml), diethylamino sulphur trifluoride (80  $\mu$ l) is added at 0°C. After stirring at 0°C for 0,5 hour, saturated sodium bicarbonate and more dichloromethane are added. The dichloromethane is dried (MgSO<sub>4</sub>), filtered and evaporated. Purification by silica gel column chromatography (ethyl acetate:hexane 1:9), gives the dissaccharide (55) in a 65% yield.

#### EXAMPLE 42

The results of the following three bioassays on the appetite suppressant are set out below, viz.

- a) Irwin Test;
- b) Acute Toxicity Test; and
  - c) Oral Dose Anorectic Test.

#### a) <u>Irwin Test</u>

The purpose of this test was to evaluate the appetite suppressant of the invention produced from a plant extract as hereinbefore described, according to the reduced animal Irwin test for tranquillising and sedative action.

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# Experimental Procedure

The appetite suppressant was extracted from plant material by the Applicant by the method as hereinbefore described and administered to two of four groups of three animals each: one group receiving no treatment, one group receiving the solvent dimethylsulfoxide (DMSO), one group receiving the test sample at 50 mg/kg, and one group receiving the test sample at 300 mg/kg. Treatment took place by intraperitoneal injection, and observations were made at specific intervals up to five hours post treatment. Only symptoms other than those observed in the DMSO-treated animals were used in the interpretation of the results.

# Results

It was clear that the solvent, DMSO, had a marked effect on the animals, especially on the heat regulating mechanism. Body temperatures of all the animals treated with the solvent, alone or together with the test sample, showed a marked drop.

Animals in the low dose group showed decreased dispersion in the cage and decreased locomotor activity, as in all the other groups, including the control group. Apathy was seen in the same degree as in the DMSO-treated group. Decreased respiration was observed 15-60 minutes after treatment. Ptosis (closing of the eyelids) was also observed to a larger degree than in the DMSO group. A

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pinna (ear) response was seen as well as a positive finger response, indicating fearfulness. Body temperature dropped to 32,7°C after treatment.

Animals in the high dose group showed as in the other groups an initial decreased dispersion in the cage and decreased locomotor activity, but showed increased dispersion and locomotor activity before death, which occurred approximately 1 hour after treatment. Severe clonic symmetrical convulsions occurred 30 minutes after treatment. Respiration decreased initially, but increased A pinna (ear) response was delayed and a before death. positive finger response was observed, indicating fearfulness, both as observed in animals in the low dose group. Body temperature dropped to 30,7°C after treatment. Increased positional passivity was observed as well as decreased body tone. Abnormal limb rotation was observed, the grip strength decreased, no pain response was present and loss of righting reflex occurred.

#### Discussion

20 When compared with the control and DMSO-treated animals, animals receiving the low dose (50 mg/kg) only showed decreased respiration and an increased degree of ptosis. Animals receiving the high dose (300 mg/kg) of the test sample reacted very intensely by showing convulsions and death. All other observations made in these animals can be ascribed to the animals being in convulsions and

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dying. Signs suggestive of tranquillising and sedative actions such as marked decreased dispersion in the cages, decreased locomotor activity and apathy in the test groups that could be ascribed to the test sample were not seen.

It can therefore be concluded that the test sample is lethal to mice at 300 mg/kg and has respiratory suppressive effects on mice at 50 mg/kg, when given intraperitoneally with DMSO as solvent.

### b) Acute Toxicity Test

The purpose of this test was to gain information on the toxicity of the test sample.

#### Experimental Procedure

A plant extract prepared in accordance with the invention as hereinbefore described, and having appetite suppressive action was purified and one test sample was tested at increasing doses by oral treatment in mice. Two animals were used per dose group, except in the highest dose group where only one animal was treated. Animals were examined for good health and their body masses determined on the day of treatment.

Doses ranged from 100 mg/kg up to 3 028,5 mg/kg. The dose was calculated and mixed into prepared potato starch, so that each animal received a total dose of 0,2 ml.

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Animal 13 received 0,25 ml. Potato starch was prepared by mixing 20 g starch into a small volume of cold water, and adding it to boiling water, to make up a volume of 1 litre. The suspension was allowed to cool to room temperature before dosing.

Animals in groups 1 and 2 were treated on the same day. They were observed for 24 hours and if no signs of toxicity developed, the next group was treated. The same approach was followed until all the animals were treated. This schedule was followed to ensure that animals were not unnecessarily treated when an acute toxic dose had been reached in the previous group.

Animals were observed for clinical signs of toxicity immediately (1-2 hours) after treatment and daily thereafter. Body mass was determined once a week and total food and water intakes of each animal were measured.

Surviving animals were euthanased by intraperitoneal injection of pentobarbitone sodium (commercially available under the trade name Euthanaze, Centaur<sup>R</sup>) on day 14 of the experiment. A post-mortem examination was performed on these animals, as well as on the one animal which died during the experiment. Samples for histopathology were collected.

#### Results

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#### Group 1 (Control Group)

No clinical signs of toxicity were observed during the 14-day observation period. Food and water intakes were within the normal parameters. Changes in body mass were also within normal parameters. No histopathological changes were recorded in the liver samples.

# Group 2 (100 mg/kg)

No clinical signs of toxicity were observed during the observation period. Food and water intakes were normal and changes in body mass over the observation period were also normal. No macroscopical pathology was observed and no histopathological or morphological changes were recorded in the liver samples.

# 15 <u>Group 3 (200 mg/kg)</u>

Animals in this group showed no clinical symptoms of toxicity during the experiment. Food and water intakes were normal, as was the change in body mass. No macroscopic pathology was observed, but the livers showed histopathological changes on examination. Cloudy swelling of the hepatocytes was mild in animal 6, but moderate in animal 5. Moderate hydropic degeneration also occurred in the hepatocytes of animal 5.

#### Group 4 (400 mg/kg)

No clinical signs of toxicity were observed during the observation period, and no macroscopic pathology was observed during the post-mortem examination. Moderate cloudy swelling and mild hydropic changes of the hepatocytes were observed on histology.

Water and food intakes and the increase in body mass in animal 7 were normal. Animal 8 consumed almost double the total food intake of animal 7 (144,6 g and 73,9 g respectively), but the increase in body mass was only 0,81 g compared to 2,7 g.

#### Group 5 (800 mg/kg)

One animal (animal 10) died three hours after dosing without showing any specific signs. The other animal (animal 9) survived the entire observation period without any signs of toxicity. Water intake in the surviving animal was normal (42,42 ml), while food intake was high (134,2 g). The body mass increased by 2,85 g which was the highest of all animals in the experiment.

At the post-mortem examination of animal 10, which died shortly after oral dosing, the lungs were congested. No foreign body reaction which would have indicated inhalation of test material was present. No macroscopic pathology was observed in animal 9. Mild cytoplasmic vacuolisation

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(hydropic degeneration) was present in animal 10, but moderate in animal 9. The glandular cytoplasmic appearance of the liver was classified as moderate in both animals.

# Group 6 (1 600 mg/kg)

5 None of the animals presented any clinical signs of toxicity during the duration of the experiment. macroscopic pathology was observed at post-mortem examination, but moderate degenerative changes in the liver animal 11 were observed histopathological at 10 examination. Animal 12 showed moderate cloudy swelling and mild hydropic changes of the hepatocytes. Food and water intakes were normal, as was the increase in body mass over the experimental period.

#### Group 7 (3 028,5 mg/kg)

Only one animal was treated at this dose. This animal showed no signs of toxicity during the observation period, and no macroscopic pathology was observed. At histopathological examination, moderate cloudy swelling and hydropic degeneration of the hepatocytes was observed. The animal showed a loss of body mass over the observation period (-0,82 g), but food and water intakes were normal.

#### Discussion

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Since a very small number of animals were used in each dose group, it is difficult to make any conclusions. The fact that only one animal died at a low dose rate, without showing any symptoms, might indicate that death was not related to the test sample, but due to stress during and/or after treatment. No animals in higher dose groups died or showed any signs of toxicity, which further supports this assumption.

The increased food intake observed in animal 8 could possibly be ascribed to excessive spillage of food as was reflected in the small increase in body mass. It should be kept in mind that all the animals in this experiment were only treated once, and that it is unlikely that an appetite suppressor will have a marked influence on either the food or water intakes, or body mass over a 14 day period, as was the case in this experiment.

From the histopathological examination of the liver samples, it was clear that the pathological changes were dose related, with animals receiving higher doses showing the extensive changes. The pathology observed was not metabolic of nature, but possibly test sample-induced. The changes were only degenerative and therefore reversible. No signs of irreversible hepatocellular changes were observed.

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It can, therefore, be concluded that only one animal died at a lower dose (800 mg/kg), but that the death was possibly not test sample related. None of the other animals in any of the dose groups showed any signs of toxicity during the 14 day observation period after treatment, or died as result of the treatment. A single oral dose of the test sample induced reversible dose-related hepatocellular changes.

# c) Oral Dose Anorectic Test

The purpose of this test was to determine the activity of a plant extract prepared in accordance with the invention, and the minimum effective dose, and at the same time investigate any possible side-effects such as respiratory suppression, as experienced in the Irwin Test (referred to above).

# Experimental Procedure

Animals were allocated to treatment groups using randomisation tables. Each treatment group consisted of three animals, with 6 animals in the control group. The test sample was dosed to young female rats with body weight 100-150 g at acclimatisation, for three consecutive days. Animals were identified by means of metallic ear tags and KMnO<sub>4</sub> skin markings for easy identification. Animals were housed individually in standard rodent polycarbonate cages, and water and powdered commercial rodent pellets were

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available <u>ad libitum</u>. Water and food intakes were measured and calculated for each day. In order to find the minimum effective dose of the test sample, five doses were tested. Treatment was by oral gavage, with the test sample suspended in potato starch.

The test substance was compound (1), a white granular powder prepared from an extract from plant material in accordance with the invention, and the measured quantity of the test sample was mixed with prepared potato starch and dosed. Mixing with potato starch took place immediately before dosing on each day. Before withdrawal of the dosing volume for each animal, the suspensions were mixed thoroughly using a Vortex.

A range of five doses was tested, with a control group receiving only the carrier substance. Doses were chosen on the basis of the effects observed in the aforedescribed Irwin Test and were:

Group 1: 0,00 mg/kg (Control Group)

Group 2: 6,25 mg/kg

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Group 4: 25,00 mg/kg

Group 5: 37,50 mg/kg

Group 6: 50,00 mg/kg

#### Results

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Treatment did not affect the health of the animals during the study period. Animals treated with the test sample in all dose groups, showed a significantly reduced mean body mass gain over the total study period, and animals in three of the five treatment groups actually lost body mass.

Mean food intakes for all the treatment groups were reduced over the study period. Animals in the higher dose groups showed an increased water consumption.

Respiratory rate in none of the animals in any dose group was significantly effected.

Animals in all dose groups presented with friable livers at post-mortem examination, but no macroscopic pathology was observed.

# Discussion

Data collected during the acclimatisation period confirmed that all animals included in the experiment were healthy and body mass gain was comparable between the animals.

The reduction, and in some animals even a loss, in body mass gain, in combination with the reduced food intake is

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strongly indicative of suppression of the appetite centre.

Reduced food intake and reduced body mass gain was experienced even with the lowest dose group (6,25 mg/kg). Actual loss in body mass was experienced in the 12,50 mg/kg group.

It is important to note that the treatment groups all had an increased water consumption when feed consumption decreased (Figure 2). This could be due to a diuretic effect of the test sample, or to stimulation of the thirst centre in the brain.

The fact that no respiratory suppression occurred as had been observed in the acute toxicity test referred to above, with the intraperitoneal route, is seen as a positive aspect. This could be due to reduced absorption from the gastrointestinal tract, with consequent reduced bioavailability. The bioavailability at the oral doses tested was, however, sufficient for the test sample to be effective. The slight reduction in respiratory rate 1 hour post treatment in most groups could be ascribed to filling of the stomach with the dose volume and consequent passivity of the animals.

The friable livers observed in the treatment groups could be due to a change in the energy metabolism secondary to the reduced food intake, causing increased fat metabolism and overload on the liver. If this was indeed

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the case, these changes could possibly be regarded these changes as transitory which might recover with time after a steady state had been reached, or after withdrawal of the test sample. The possible effect on the liver also needs further investigation.

Since this study was intended primarily as a screening test, small groups of test animals were used. This makes interpretation statistical of the data difficult, individual where animals react especially totally differently. However, the data indicates that the test sample has appetite suppressive action, even at the lowest dose tested (6,25 mg/kg). No clinical signs of respiratory suppression occurred at the doses tested.

#### EXAMPLE 43

Harvested Hoodia plants received either from the natural environment or through a cultivation programme are first stored at 4°C for a maximum of 48 hours. The plants are washed in tap water and thereafter sliced into ± 1 cm slices. The sliced pieces are all combined and then pressed through a hydraulic press at 300 bar pressure for a minimum of 0.5 hour per pressing. During the pressing the sap of the plant is collected separately. The sap is stored at -18°C until further processing is required.

The sap is spray-dried under suitable conditions to obtain a free flowing powder. The moisture content in the powder

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is preferably less than 5% after spray drying and, if necessary, it is further dried in a vacuum oven or using a fluid bed drier.

Both the sap and the spray-dried material have been shown effective as an appetite suppressant in biological assays in rats.

#### Experimental

50 kg of Hoodia gordonii plants were washed with tap water and thereafter sliced into 1 cm slices. The sliced plants were then pressed through a hydraulic press at 300 bar for a minimum of 0.5 hour per batch. The sap was collected and the mass was found to be 10 kg when Hoodia gordonii plants from the environment were used, and 20 kg when Hoodia gordonii plants from the cultivation programme was used. The sap (500 g) was spray-dried using the following conditions:

Flow rate : 2.85 ml/min

Inlet temperature : 110°C
Outlet temperature : 70°C
Chamber temperature : 78°C

The spray-dried powder obtained was a free flowing powder (22 g) with a moisture content of 6.9%.

The spray dried powder was analysed for active ingredient concentration using HPLC techniques. The concentration of the active was determined to be 13 g/kg of spray dried powder.

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#### HPLC Analysis Method

Eluant : Acetonitrile: water (7:3),

isocratic

Column : Reverse phase C-18

UV absorbance : 225 nm Flow rate : 1 ml/min Injection volume :  $10\mu$ l

#### <u>Method</u>

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Spray-dried powder (10 mg) was dissolved in water (0.5 ml) and acetonitrile (0.5 ml)  $10\mu$ l of this solution was injected into the HPLC and the concentration of the active compound (1) was determined using a standard curve which was prepared from the pure compound (1).

#### EXAMPLE 44

The results of a study designed to assess the possible anorectic effects of compound (1) in the rat are presented below. In the following, the samples tested are pure sap (Sample 1), spray-dried sap (Sample 2) and active moiety (Sample 3). Samples 1 and 2 are the sap and the spray-dried sap respectively, as described in Example 43 above. Sample 3 is solvent-extracted compound (1) of ≥95% purity.

Sample 1 to 3 were each administered as a single oral dose to male Wistar rats. Two additional control groups received vehicle (distilled water or DMSO). Orally administered fenfluramine (7.5 mg/kg) was included as a reference standard.

Sample 1 (pure sap) administered orally, produced dose-dependent reductions in food consumption which were statistically significant at doses of 1600 mg/kg and above when compared with vehicle-treated controls. Concomitant reductions in bodyweight (or growth rate) were also recorded. On the day of dosing, statistically significant

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increases in water consumption were recorded at 3 hours post-dose (6400 and 10000 mg/kg) and 6 hours post-dose (10000 mg/kg). Between 24 and 48 hours post-dose, statistically significant reductions in water consumption were recorded at doses of 3200 mg/kg and above.

Sample 2 (spray-dried sap) administered orally at 76 mg/kg also produced statistically significant reductions in food consumption and bodyweight when compared with vehicle-treated animals. No statistically significant effects on water consumption were recorded.

Sample 3 (active moiety) produced statistically significant reductions in food consumption at an oral dose of 5.0 mg/kg. No statistically significant effects on bodyweights were produced by the active moiety although examination of the data revealed a slight delay in growth when compared with vehicle-treated control animals. No statistically significant effects on water consumption were recorded.

The reference standard, fenfluramine (7.5 mg/kg), produced statistically significant reductions in food consumption at 6 and 24 hours post-dose when compared with the relevant vehicle-treated control group. No statistically significant effects on water consumption or bodyweight were recorded.

No treatment-related effects on the livers were recorded.

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Identity .	Sample 1 (pure sap)	Sample 2 (spray-dried sap)	Sample 3 (active molecy)
Appearance B	Brown liquid	Powder	White powder
nditions	-20°C in the dark	Room temperature in the	4°C in the dark
		dark	
Purity	Pure sap	Pure spray-dried sap	835% 33
Vehicle	Distilled water	Distilled water	Dimethylsulphoxide (DMSO)

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#### Experimental Procedure

Fifty-five male Wistar rats were used for the study.

Bodyweights, food consumption (food hopper weight) and water consumption (bottle weight) were recorded daily at the same time each day from the day of arrival until the termination of the study.

On Day 1, the rats received a single oral (gavage) dose according to the following table:

	Group	n	Oral treatment	Dose (mg/kg)
10	1	5	Vehicle (distilled water)	-
	2	4	Sample 1 (pure sap)	800
	3	5	Sample 1 (pure sap)	1600
	4	5	Sample 1 (pure sap)	3200
	5	5	Sample 1 (pure sap)	6400
15	6	5	Sample 1 (pure sap)	10000
	7	5	Sample 2 spray-dried sap	38
	8	5	Sample 2 spray-dried sap	76
	9	5	Sample 3 (active moiety)	2.5
	10	5	Sample 3 (active moiety)	5.0
20	11	3	Fenfluramine	7.5
	12	3	Vehicle (DMSO)	-

Groups 1 - 8 were dosed using a constant dose volume of 10 ml/kg and groups 9 - 12 were dosed using a dose volume of 1 ml/kg.

25 Food and water consumption were also measured at 1,3 and 6 hours after dosing on Day 1.

Following the measurements of Day 8, the animals were killed by carbon dioxide asphyxiation, and the livers excised and placed in 10% buffered formalin, prior to histology.

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Paraffin wax sections of each liver were taken at 4 - 5  $\mu m$  and stained with haematoxylin and eosin. Additional sections were cut on a cryostat at 12  $\mu m$  and stained for fat with Oil Red O (ORO).

# 5 <u>Data Analysis</u>

The post-dose food and water consumption measurements and bodyweights at each time-point for the P57-treated animals were compared with those for the relevant, similarly-treated vehicle control group using analysis of variance followed by Williams' test for comparisons with controls.

The data for the fenfluramine-treated animals was compared with that for the vehicle-treated control group using Student's t test.

#### Results

The results are summarised in the tables.

Sample 1 (pure sap) administered orally produced marked, dose-related reductions in daily food consumption. The duration and amplitude of these reductions in food consumption were dose-dependent. At 24 hours post-dose, Sample 1 (pure sap) produced statistically significant reductions in food consumption at doses of 1600 mg/kg and above when compared with vehicle-treated controls. The highest dose of Sample 1 (sap) (10000 mg/kg) produced statistically significant reductions in food consumption on a daily basis up to 5 days post-dose.

Sample 2 (spray-dried sap) and Sample 3 (active moiety) produced marked and statistically significant reductions in food consumption at oral doses of 76 and 5.0 mg/kg respectively. In both cases the effects lasted 48 hours post-dose.

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The reference standard, fenfluramine (7.5 mg/kg, p.o.) produced statistically significant reductions in food consumption at 6 and 24 hours post-dose when compared with the relevant vehicle-treated control group (Group 12).

Sample 2 (spray-dried sap) and Sample 3 (active moiety) marked, dose-related produced no effects On the day of dosing, the pure sap produced consumption. statistically significant increases in water consumption at 3 hours post-dose (6400 and 10000 mg/kg) and 6 hours post-Two days after dosing however, (10000 mg/kg). statistically significant decreases in water consumption were recorded in animals receiving Sample 1 (sap) at 3200, These reductions however, were not 6400 and 10000 mg/kg. clearly dose-related and only occurred between 1 and 2 days The biological significance of these effects post-dose. therefore remains unclear.

Sample 1 (pure sap) produced dose-related, statistically significant effects on bodyweights when compared with the vehicle-treated control group (Group 1). When administered orally at doses of 3200 mg/kg and above, Sample 1 (pure sap) produced statistically significant reductions in bodyweight or decreased growth rates when compared with vehicle-treated animals. These effects were statistically significant from 48 hours post-dose until the end of the study.

Sample 2 (spray-dried sap) administered orally at 76 mg/kg also produced statistically significant reductions in growth of the animals when compared with the vehicle-treated control group (Group 1). These effects were statistically significant between Days 3 (48 hours postdose) and 5 inclusive.

Although Sample 3 (active moiety) appeared to delay the growth of the animals at the highest dose (5.0 mg/kg) when compared with the relevant vehicle-treated control group

(Group 12), this effect was not statistically significant.

Fenfluramine, (7.5 mg/kg) produced no marked or statistically significant effects on water consumption or bodyweights when compared with the vehicle-treated control group (Group 12).

No treatment-related effects on the livers were recorded.

TABLE 1a Effects of oral administration on food consumption in the rat (daily pre-dose data)

Group	Oral treatment	Dose (mg/kg)	Group mea	n food cons	umption (g	Group mean food consumption (g ± sd) between Days:	reen Days:
		,	- 9 - - 9 -	- 5 -	-4-	-5-	-2-
-	Vehicle (water)	•	27.8 ±1.54	24.2 ±1.83	27.6 ±3.67	28.3 ±3.50	29.4 ±2.66
2	Sample 1 sap	800	28.3 ±1.43	24.9 ±0.82	27.7 ±0.76	28.4 ±1.51	30.1 ±0.27
3	Sample 1 sap	1600	29.0 ±1.39	25.0 ±2.16	27.4 ±1.96	28.8 ±0.61	29.5 ±1.55
4	Sample 1 sap	3200	27.2 ±2.33	25.1 ±2.46	26.0 ±2.52	28.5 ±2.29	27.6 ±1.15
2	Sample 1 sap	6400	28.7 ±1.64	25.3 ±1.73	27.3 ±1.45	29.2 ±1.09	30.3 ±0.90
9	Sample 1 sap	10000	28.5 ±2.38	23.7 ±2.73	26.0 ±2.31	27.0 ±3.50	28.7 ±2.26
7	Sample 2 spray-dried	38	28.1 ±1.24	23.9 ±1.79	24.5 ±2.30	27.6 ±1.61	28.5 ±1.87
8	Sample 2 spray-dried	76	28.7 ±0.91	26.5 ±1.55	27.1 ±1.01	28.7 ±1.99	28.9 ±1.37
6	Sample 3 active molety	2.5	28.8 ±1.49	26.4 ±3.12	29.0 ±1.99	29.4 ±1.76	29.5 ±2.81
10	Sample 3 active molety	5.0	28.3 ±2.1	25.8 ±1.86	28.1 ±2.65	28.0 ±2.65	28.5 ±3.03
11	Fenfluramine	7.5	29.1 ±0.66	25.3 ±4.03	27.0 ±1.53	30.8 ±0.54	29.7 ±2.84
.12	Vehicle (DMSO)	•	27.9 ±1.8	26.7 ±2.11	28.7 ±1.99	28.1 ±4.06	30.5 ±2.54

sd Standard deviation

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TABLE 1b Effects of oral administration on food consumption in the rat (daily post-dose data)

Groun	Oral treatment	Dose (ma/kg)		Group mea	n food cons	Group mean food consumption (g	± sd) between Days:	ween Day	S:
		•	1-2	2-3	3 - 4	4 - 5	5 - 6	6-7	7 - 8
-	Vehicle (water)	•	29.5 ±3.15	29.6 ±2.84	30.6 ±3.49	31.8 ±3.21	30.7 ±2.24	31.7 ±3.03	32.9 ±3.18
2	Sample 1 sap	800	26.1 ±0.98	29.3 ±1.49	30.7 ±1.15	30.9 ±0.60	33.3 ±1.69	32.7 ±0.80	40.1 ±13.40
က	Sample 1 sap	1600	22.6 ±3.17	26.9 ±2.06	30.9 ±2.54	30.9 ±1.22	34.1 ±1.36	33.7 ±1.69	33.8 ±1.61
4	Sample 1 sap	3200	20.1 ±1.39	19.0 ±1.88	22.8 ±1.77	28.0 ±3.14	31.4 ±2.82	32.3 ±2.91	33.0 ±3.01
2	Sample 1 sap	6400	18.2 ±4.18	14.8 ±1.75	18.4 ±0.97	22.4 ±3.01	26.9 ±2.81	31.0 ±2.31	32.0 ±2.34
9	Sample 1 sap	10000	15.1 ±2.98	12.4** ±2.61	16.0 ±3.15	19.7 ±4.31	22.6 ±5.70	30.1 ±4.79	32.6 ±5.90
7	Sample 2 spray-dried	38	25.6 ±2.85	27.3 ±0.95	30.3 ±2.06	31.0 ±2.13	31.8 ±1.63	31.1 ±1.94	31.8 ±2.45
8	Sample 2 spray-dried	76	24.2 ±3.25	25.2* ±3.24	29.9 ±1.85	30.2 ±2.28	31.2 ±2.26	32.3 ±1.44	33.1 ±0.61
6	Sample 3 active molety	2.5	26.8 ±3.33	29.1 ±3.43	31.7 ±3.08	34.0 ±2.95	34.4 ±4.32	33.1 ±4.11	34.8 ±3.71
10	Sample 3 active molety	5.0	22.1 <sup>††</sup> ±2.19	21.0 <sup>††</sup> ±3.07	27.6 ±5.26	30.5 ±3.33	33.0 ±3.16	32.4 ±3.25	33.0 ±3.84
11	Fenfluramine	7.5	22.4 <sup>†</sup> ±3.19	31.9 ±0.84	32.7 ±2.50	33.0 ±2.55	30.4 ±0.23	32.7 ±1.90	32.4 ±1.60
12	Vehicle (DMSO)	•	29.9 ±3.36	30.6 ±4.43	30.1 ±4.17	32.4 ±5.26	31.8 ±3.08	32.8 ±3.98	33.3 ±3.76

sd Standard deviation Group 1: p < 0.05, p < 0.01 Groups 2 - 8 were compared with vehicle Group 12: p < 0.05, p < 0.01 Groups 9 - 11 were compared with vehicle Group 12: p < 0.05,

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Fffe	Effects of oral administration on water consumption in the rat (daily pre-dose data)	on on water o	sonsumptic	ption in the r	at (daily p	re-dose da	ita)
Group	Oral treatment	Dose (mg/kg)	Group mea	n water cor	Group mean water consumption (g ± sd) between Days:	± sd) bet	ween Days:
200			- 6 - -5	-5-	-4-	-3-	-2-
-	Vehicle (water)		40.9 ±4.61	34.8 ±4.15	37.6 ±5.63	33.5 ±7.42	32.2 ±6.32
2	Sample 1 sap	800	38.6 ±1.96	37.1 ±9.74	36.4 ±4.81	28.1 ±1.83	30.4 ±4.75
3	Sample 1 sap	1600	43.4 ±10.53	35.9 ±3.84	38.4 ±4.56	31.1 ±4.47	36.5 ±5.39
4	Sample 1 sap	3200	40.1 ±5.58	33.3 ±3.01	37.3 ±4.46	31.3 ±3.48	31.7 ±3.18
2	Sample 1 sap	6400	43.8 ±8.57	36.3 ±9.02	35.4 ±8.18	34.0 ±6.62	35.1 ±5.72
9	Sample 1 sap	10000	37.4 ±5.34	32.7 ±3.35	33.2 ±4.86	29.0 ±5.11	32.2 ±3.27
7	Sample 2 spray-dried	38	40.0 ±4.36	35.8 ±4.92	34.7 ±3.20	30.2 ±1.88	31.4 ±2.98
8	Sample 2 spray-dried	76	38.6 ±1.98	37.0 ±1.96	48.8 ±21.5	31.6 ±4.56	39.0 ±17.27
6	Sample 3 active molety	2.5	42.0 ±6.70	37.0 ±5.05	34.1 ±3.16	28.0 ±2.58	31.6 ±3.12
10	Sample 3 active molety	5.0	40.9 ±4.48	34.2 ±3.00	32.7 ±1.26	28.2 ±1.65	33.1 ±4.82
11	Fenfluramine	7.5	47.0 ±5.3	35.5 ±7.49	34.7 ±3.73	30.9 ±2.12	31.6 ±2.80
12	Vehicle (DMSO)		43.3 ±5.67	34.5 ±4.97	35.2 ±4.34	28.3 ±4.64	31.4 ±6.44

sd Standard deviation

Effects of oral administration on water consumption in the rat (daily post-dose data) TABLE 2b

Group	Oral treatment	Dose (mg/kg)		Group mear	Group mean water consumption (g	sumption (c		± sd) between Days:	
			1-2	2-3	3 - 4	4 - 5	5 - 8	6 - 7	7 - 8
-	Vehicle (water)	,	34.9 ±5.45	36.9 ±6.06	38.0 ±7.59	37.2 ±6.18	37.7 ±5.54	35.3 ±2.86	36.5 ±5.85
2	Sample 1 sap	800	30.9 ±3.77	34.4 ±8.12	38.2 ±13.71	35.9 ±13.51	39.5 ±11.20	28.8 ±1.22	31.8 ±5.58
က	Sample 1 sap	1600	29.2 ±1.66	31.7 ±5.35	41.3 ±11.21	34.6 ±4.10	48.1 ±12.27	37.8 ±7.28	36.9 ±9.28
4	Sample 1 sap	3200	35.9 ±5.88	26.2 ±2.66	30.5 ±2.44	34.1 ±4.80	45.8 ±18.54	51.0 ±35.21	42.6 ±13.88
S.	Sample 1 sap	6400	33.4 ±12.04	27.4 ±8.13	32.6 ±10.67	35.4 ±10.78	45.2 ±8.72	36.2 ±6.72	35.9 ±9.58
ဖ	Sample 1 sap	10000	31.7 ±12.74	28.5 ±8.85	32.4 ±8.87	36.6 ±6.50	40.7 ±11.51	38.0 ±6.66	37.5 ±6.21
7	Sample 2 spray-dried	38	36.0 ±6.02	34.5 ±1.79	38.2 ±7.16	39.6 ±7.09	42.7 ±9.74	45.6 ±17.15	46.1 ±9.49
8	Sample 2 spray-dried	76	45.0 ±19.03	39.1 ±16.59	46.9 ±18.34	35.9 ±3.40	41.9 ±12.37	36.9 ±8.47	38.1 ±8.93
6	Sample 3 active molety	2.5	32.2 ±4.01	36.1 ±12.42	38.3 ±11.71	41.5 ±16.60	34.7 ±7.57	33.0 ±4.20	35.3 ±8.70
10	Sample 3 active molety	5.0	33.9 ±2.40	31.5 ±8.12	35.1 ±3.82	37.7 ±5.99	39.5 ±7.78	37.4 ±11.07	37.8 ±6.42
=	Fenfluramine	7.5	34.1 ±3.60	37.2 ±1.48	36.7 ±3.92	33.8 ±2.89	33.7 ±5.43	32.1 ±1.93	33.6 ±2.50
12	Vehicle (DMSO)	•	40.7 ±9.10	33.8 ±9.37	32.9 ±7.07	35.2 ±11.49	33.8 ±9.82	32.3 ±7.44	32.0 ±7.22

sd Standard deviation Group 1:  $^*\rho$  < 0.05 Groups 2 - 8 were compared with vehicle Group 1:  $^*\rho$  < 0.05 Groups 9 - 11 were compared with vehicle Group 12 (no significances)

Effect	I ABLE 3a Effects of oral administration on bodyweight in the rat (daily pre-dose data)	on bodyweig	ABLE 3a tht in the	rat (dail)	pre-dos	e data)	
2012	Oral treatment	Dose (ma/ka)	Group me	Group mean bodyweight (g	eight (g ±	sd on Day:	
dooio			-5	-4	6	-2	1-
-	Vehicle (water)	t.	130.9 ±5.56	150.7 ±5.37	157.3 ±5.29	168.1 ±6.20	177.5 ±6.70
2	Sample 1 sap	800	131.6 ±4.34	150.1 ±4.84	158.5 ±4.35	169.6 ±4.99	177.7 ±4.10
3	Sample 1 sap	1600	130.1 ±4.3	148.6 ±6.59	156.7 ±6.38	167.5 ±6.04	176.6 ±6.37
4	Sample 1 sap	3200	130.8 ±6.19	147.7 ±7.56	154.4 ±8.06	165.2 ±8.43	175.8 ±9.10
2	Sample 1 sap	6400	132.6 ±7.01	151.3 ±7.23	158.4 ±8.50	169.0 ±8.79	178.1 ±7.75
9	Sample 1 sap	10000	132.3 ±6.75	151.8 ±9.08	157.3 ±9.37	167.1 ±10.41	175.4 ±10.90
_	Sample 2 spray-dried	38	131.7 ±8.28	149.0 ±5.85	156.2 ±5.81	166.7 ±5.54	175.6 ±8.42
8	Sample 2 spray-dried	76	130.0 ±6.99	146.1 ±6.00	155.9 ±6.59	166.0 ±6.87	175.1 ±6.55
6	Sample 3 active molety	2.5	132.6 ±7.63	148.9 ±8.51	157.3 ±8.91	169.8 ±8.96	179.4 ±8.71
10	Sample 3 active molety	5.0	133.5 ±6.45	150.5 ±9.55	158.8 ±8.48	171.0 ±7.72	179.0 ±9.20
11	Fenfluramine	7.5	133.2 ±9.21	152.7 ±9.09	160.0 ±9.82	170.0 ±9.15	182.8 ±10.21
12	Vehicle (DMSO)	•	129.1 ±3.17	147.3 ±4.37	155.0 ±6.29	166.0 ±5.91	174.8 ±8.26

sd Standard deviation

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Effects of oral administration on bodyweight in the rat (daily post-dose data)

Group	Oral treatment	Dose (mg/kg)			Group me	Group mean bodyweight (g ± sd) on Day:	ht (g ± sd) o	n Day:		
			Pre-dose (1)	2	3	4	5	9	7	8
-	Vehicle (water)	•	185.4 ±7.77	192.6 ±7.16	202.0 ±10.17	211.2 ±7.98	220.2 ±10.35	227.2 ±10.26	235.8 ±11.82	242.8 ±11.97
2	Sample 1 sap	800	186.0 ±4.90	187.0 ±4.55	198.5 ±4.20	206.8 ±5.91	214.8 ±4.65	222.8 ± 4.99	231.5 ±3.70	240.0 ±3.65
е	Sample 1 sap	1600	185.0 ±6.67	186.0 ±8.28	193.2 ±6.42	204.0 ± 6.40	212.4 ±5.81	223.0 ±6.33	232.6 ±7.70	240.4 ±6.66
4	Sample 1 sap	3200	181.8 ±9.18	184.6 ±8.88	186.2 ±8.67	189.8* ±9.99	199.2** ±9.34	210.6 ±10.21	219.0" ±11.29	228.4 ±12.18
2	Semple 1 sep	6400	186.6 ±7.96	185.6` ±6.39	183.8 ±6.87	185.2 ± ± 9.18	191.2** ±7.89	201.0 ±6.89	213.0 ±6.96	222.0" ±7.94
9	Sample 1 sap	10000	182.8 ±12.22	181.4 ±14.06	179.8 ±15.85	180.6 ±13.85	185.6 ±11.28	192.2 ±10.99	203.4 ±11.68	212.4 ±11.35
7	Sample 2 spray-dried	38	183.4 ±8.11	185.8 ±9.23	195.8 ±7.79	205.6 ±9.79	214.4 ±9.61	222.6 ±9.34	231.4 ±10.62	239.6 ±11.46
В	Sample 2 spray-dried	76	180.6 ±6.47	183.4 ±7.57	188.6 ±6.73	198.2 ±8.50	206.0 ±9.43	214.0 ±9.51	222.0 ±9.49	232.2 ±9.68
6	Sample 3 active molety	2.5	188.2 ±9.42	191.2 ±11.15	200.0 ±11.25	209.6 ±12.28	219.6 ±12.95	229.4 ±13.69	238.4 ±14.50	247.0 ±14.35
10	Sample 3 active molety	5.0	186.4 ±10.02	192.0 ±9.93	192.4 ±9.84	201.0 ±11.27	209.4 ±12.70	219.8 ±11.86	228.2 ±12.28	236.0 ±13.95
11	Fenfluramine	7.5	190.3 ±9.71	190.3 ±10.97	197.7 ±7.37	207.7 ±7.23	217.7 ±10.69	224.3 ±10.12	234.3 ±12.70	243.3 ±9.24
12	Velvicie (DMSO)	•	183.3 ±8.33	190.3 ±10.26	199.0 ±10.82	207.7 ±12.66	215.7 ±14.05	222.3 ±14.84	230.7 ±15.95	239.0 ±17.35

sd Stendard deviation Groups 2 - 8 were compared with vehicle Group 1:  $^{\bullet}\rho$ <0.05,  $^{\bullet}\rho$ <0.01 Groups 9 - 1:1 were compared with vehicle Group 12 (no significances)

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# Histopathology Report

Histological examination was restricted to the liver. No treatment-related changes were detected for Sample 1 (liquid), Sample 2 (spray-dried sap), Sample 3 (active moiety), fenfluramine or the DMSO control group.

The findings recorded were of a similar incidence in control and treated groups.

TABLE

Microscopic pathology incidence summary

	Group 1	Group 2	Group 3	Group 4	Group 5	Group 6
Sex: Males	o mg/kg	800 mg/kg	1600 mg/kg	3200 mg/kg	6400 mg/kg	10000 mg/kg
Males on study Animals completed	សស	কক	5	សស	សហ	N N
Examined No abnormalities detected No abnormalities detected Minimal Hepatcorte hypertrophy - centrilobular (Total) Minimal Extramedullary haemopoiesis (Total Minimal Hepatcorte necrosis - focal (Total) Minimal Hepatcoyte necrosis - focal (Total) Minimal Portal lymphoid infiltration (Total) Minimal Eosinophilic hepatcoytes - focal (Total) Minimal Portal lymphoid infiltration (Total) Minimal Minimal Portal fibrosis (Total)		401100000440000	N H O O O O O O O O 4 4 O O H H	N N O O O O O O O M M O O O O	00000000000000000000000000000000000000	wommuч0000000000
Liver (ORO stain) Examined No abnormalities detected Hepatocyte fat - centrilobular (Total) Minimal Hepatocyte fat - periportal (Total) Minimal	20 M M M D	464400	112225	87 <b>4</b> 11 11 0 0	5 F C C C C C	N W N N O O

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TABLE (continued)

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Group 12 0 mg/kg	т п	wa44004400;	6710
Group 11 7.5 mg/kg	m m	m 0 0 0 0 0 m m 0 0	манн
Group 10 5 mg/kg	សស	N 4 0 0 0 0 4 4 0 0	2 2 2 2 D
Group 9 2.5 mg/kg	ດາເນ	поооннимны	N 12 12 12
Group 8 76 mg/kg	'nΜ	wwaaaawwaa	សមសស
Group 7 38 mg/kg	ហហ	w w a a a a a a a a	v v o o
Sex: Males	Males on study Animals completed	Liver Examined No abnormalities detected No abnormalities detected Parenchymal inflammatory cell foci (Total) Minimal Hepatocyte necrosis - focal (Total) Minimal Portal lymphoid infiltration (Total) Minimal Portal leucocytes (Total) Minimal Minimal	Liver (ORO stain) Examined No abnormalities detected Hepatocyte fat - centrilobular (Total) Minimal

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#### EXAMPLE 45

A further bioassay, which employed the same test samples as described in Example 44, is described below. Animals in this study received a restricted diet i.e. animals only received food between 12:00 and 3:00pm daily. different from all other biological assays conducted thus far, whereby food was available to the rats at lib. Animals were acclimatised over a seven day period (days -7 to -1), dosing took place from day 0 to day 6 at 9:00am by The recovery period was from days 7 to day oral gavage. Dosage groups are described in Table 1 below. should be noted that the actual control group is labelled Group 09. Group 5 is a controlled group which received a diet equivalent to that of Group 4. The purpose of this group was to evaluate the effect a restricted diet has on the lives of the animals.

#### Results

The results generated during the study showed that the acclimatization period was too short. Rats feed mainly during the night and the sudden change to a restricted access to feed for 3 hours during day-time, resulted in low daily intakes. The daily intake of feed was still increasing in most groups at the end of the acclimatization period when dosing with the test items started. As a result of this, the effect of the test materials did not significantly affect the food intake of the rats during the period of dosing.

The mean body masses for the different groups for day -7 to -1 and days 0 to 6 are shown in the Table D1 and Table D2. The effect of the different dosages of the sap and spray-dried sap is shown in the accompanying graphs as % change in body mass day 0 to 7 (Figure 5), and % change in body mass day -7 to 7 (Figure 6). The loss in body mass is clearly dose-related especially with the higher dosages.

The histopathological examination of the livers did not show any significant pathology in the groups receiving the test items.

#### Food

Food consumption was measured daily, during acclimatization and during the study. Food was available for a 3 hour feeding period daily, starting at 12:00 and ending at 15:00. The animals were fasted for the remainder of the time. Animals in Group 5 received a measured quantity food on Day 1, equivalent to the average food consumption of Group 4 on Day 0. This controlled feeding pattern for Group 5, as determined from the average food consumption of Group 4 from the previous day, was followed for Days 1 - 7.

#### Water

Water was provided in standard containers. Water (Magalies Water Board Tap Water, suitable for human consumption) was available ad libitum. Water consumption was measured once daily, at the same time each day, after food consumption determination.

# 20 Acclimatization

The animals were acclimatized for seven days before the start of the study, during which time food and water consumption were determined as described above. The body masses were determined on a daily basis during this time.

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# Study Design and Procedures

#### TABLE 1 STUDY DESIGN

GROUP	TEST	NUMBERS	DOSE	TEST ITEM
01	රේ	001 - 006	100 mg/kg	Frozen sap
02	6ರೆ	007 - 012	400 mg/kg	Frozen sap
03	රේ	013 - 018	1 600 mg/kg	Frozen sap
04	6ರೆ	019 - 024	3 200 mg/kg	Frozen sap
05	6ඊ	025 - 030	CONTROL	Elga Option 4 Purified Water
06	රේ	031 - 036	2.2 mg/kg	Spray-dried sap
07	රේ	037 - 042	8.8 mg/kg	Spray-dried sap
08	රේ	043 - 048	35 mg/kg	Spray-dried sap
09	රේ	049 -054	CONTROL	Elga Option 4 Purified Water

# Route of Administration

The test items were administered on a daily basis for seven days, using an intra-gastric needle. Animals were fasted for 18 hours prior to the item administration (starting at 09:00).

# Duration of Treatment

Animals were treated for seven consecutive days (from Day 0 - Day 6). Three animals of each group were sacrificed 24 hours after the last dosing (Day 7). The remaining three animals were sacrificed 7 days after the last treatment (Day 13). This procedure was followed for all the groups except for Group 5 where three animals were sacrificed 24 hours after the last controlled feeding (Day 8), the remaining three animals were sacrificed 7 days after the last treatment (Day 13).

#### Body Masses

Body masses were determined daily, at approximately the same time each day for the duration of the study, including during the acclimatization period.

# 5 <u>Euthanasia</u>

Three animals of each group were sacrificed 24 hours after the last dosing (Day 7).

The remaining three animals were sacrificed 7 days after the last treatment. This procedure was followed for all the groups except for Group 5 where three animals were sacrificed 24 hours after the last controlled feeding (Day 8), the remaining three animals were sacrificed 7 Days after the last treatment (Day 13). The animals were euthanased at the end of the study period with  $\mathrm{CO}_2$  gas.

# 15 Ophthalmoscopic Examinations

Ophthalmoscopic examinations, using an ophthalmoscope, were done prior to the first adminstration of the test item and at termination, in all animals in all groups.

# Macroscopic Pathology

A full post mortem examination was performed on every animal which was euthanased at the end of the study period.

#### <u>Histopathology</u>

Histopathological examination was performed on the liver of each of the animals.

TABLE D.1 MEAN BODY MASSES / GROUP / WEEK

Group	Oral treatment	Dose		Mean	body mass	Mean body masses (g) & Standard deviation	ındard devia	ıtion	
		(mg/kg)	Day -7	Day -6	Day -5	Day -4	Day -3	Day -2	Day -1
9	Sample 1 (Sap)	100	203.38 ±95.39	197.13 ±90.63	192.75 ±89.49	188.62 ±86.75	184.95 ±84.80	182.48 ±83.47	182.25 ±82.57
02	Sample 1 (Sap)	400	192.53 ±65.60	183.92 ±61.20	178.25 ±59.37	173.17 ±58.10	170.82 ±57.42	168.25 ±58.40	169.37 ±59.25
03	Sample 1 (Sap)	1 600	149.25 ±54.80	142.87 ±51.89	136.85 ±52.17	132.37 ±49.64	131.50 ±49.50	129.67 ±48.89	131.12 ±48.22
04	Sample 1 (Sap)	3 200	224.15 ±80.70	214.45 ±77.25	207.10 ±76.38	201.82 ±75.42	198.25 ±74.82	194.83 ±75.34	196.77 ±74.56
90	Elga Option 4 purified water (control)	•	214.55 ±74.90	204.85 ±72.41	198.57 ±71.79	193.48 ±68.49	192.40 ±67.48	190.87 ±67.39	190.15 ±65.24
90	Sample 2 (Spray-dried sap)	2.2	208.65 ±65.74	199.37 ±62.49	193.18 ±61.18	188.25 ±60.89	186.22 ±59.98	184.55 ±58.86	185.97 ±58.76
07	Sample 2 (Spray-dried sap)	8.8	256.95 ±77.55	246.02 ±73.67	237.47 ±73.53	232.62 ±71.73	229.78 ±71.76	228.07 ±69.88	228.45 ±68.81
80	Sample 2 (Spray-dried sap)	35	194.37 ±43.74	185.83 ±42.70	177.53 ±41.10	172.05 ±40.13	170.10 ±39.49	167.25 ±37.61	168.00 ±38.83
60	Elga Option 4 purified water (control)	. •	171.52 ±69.81	162.67 ±62.68	154.95 ±61.83	151.38 ±59.48	149.63 ±57.66	148.30 ±57.12	149.07 ±56.01

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TABLE D.2 MEAN BODY MASSES / GROUP / WEEK (CONTINUED)

Group	Oral treatment	Dose		Mear	Mean body masses (g) & Standard deviation	es (g) & Str	andard devis	ıtion	
		(mg/kg)	Day 0	Day 1	Day 2	Day 3	Day 4	Day 5	Day 6
.10	Sample 1 (Sap)	100	183.87 ±83.33	175.83 ±81.82	175.72 ±79.05	175.48 ±77.54	175.53 ±76.20	177.95 ±73.99	178.43 ±72.68
02	Sample 1 (Sap)	400	173.45 ±60.73	164,58 ±58.52	164.75 ±58.37	166.22 ±57.69	166.55 ±57.79	169.93 ±57.47	171.77 ±57.29
03	Sample (Sap)	1 600	134.38 ±46.01	129.20 ±44.74	127.53 ±43.20	127.20 ±41.36	126.70 ±39.19	128.00 ±39.22	128.07 ±38.66
40	Sample (Sap)	3 200	199.60 ±75.16	196.38 ±73.96	192.20 ±71.20	189.05 ±69.11	186.57 ±66.29	186.05 ±67.45	185.68 ±65.73
. 50	Elga Option 4 purified water (control)	,	194.27 ±67.46	187.93 ±65.48	181.97 ±65.01	177.53 ±64.73	174.73 ±61.08	172.85 ±58.63	171.45 ±56.79
90	Sample 2 (Spray-dried sap)	2.2	189.07 ±60.15	181.52 ±58.99	181.48 ±57.79	184.42 ±55.64	185.75 ±55.29	189.35 ±54.66	189.68 ±53.70
07	Sample 2 (Spray-dried sap)	8.8	230.28 ±69.32	221.55 ±68.02	220.17 ±66.63	221.80 ±63.88	222.82 ±63.56	224.82 ±62.38	224.90 ±62.05
88	Sample 2 (Spray-dried sap)	35	169.10 ±38.40	164.42 ±38.03	162.50 ±36.81	162.75 ±36.36	162.52 ±36.93	164.30 ±37.69	164.22 ±37.18
60	Elga Option 4 purified water (control)	•	151.02 ±55.45	146.55 ±53.77	148.10 ±52.67	149.70 ±52.05	152.58 ±50.37	155.82 ±49.91	157.85 ±49.70

TABLE D.3 MEAN BODY MASSES / GROUP / WEEK (CONTINUED)

a rose	Oral treatment	Dose		Mea	Mean body masses (g) & Standard deviation	es (g) & Star	ndard devia	tion	
3		(mg/kg)	Day 7	Day 8	Day 9	Day 10	Day 11	Day 12	Day 13
10	Sample 1 (Sap) (GHA I 35A)	100	185.38 ±72.64	234.73 ±62.44	236.73 ±62.39	234.07 ±62.09	236.33 ±62.31	239.07 ±60.24	238.43 ±59.85
02	Sample 1 (Sap) (GHA I 35A)	400	178.83 ±58.24	225.63 ±13.05	277.13 ±14.18	227.10 ±14.03	229.43 ±16.97	234.93 ±18.35	236.20 ±15.97
03	Sample 1 (Sap) (GHA I 35A)	1 600	132.22 ±37.08	133.80 ±55.17	135.23 ±455.74	134.53 ±54.96	138.30 ±53.03	139.30 ±51.10	142.80 ±49.51
40	Sample 1 (Sap) (GHA 9 35A)	3 200	188.57 ±66.14	199.63 ±61.07	198.90 ±57.48	198.70 ±54.55	194.73 ±52.78	194.93 ±50.78	197.93 ±51.57
05	Elga Option 4 purifled water (control)	•	173.97 ±54.29	172.98 ±52.06	157.80 ±58.62	158.87 ±57.76	160.80 ±57.67	163.40 ±56.27	167.80 ±58.49
90	Sample 2 (Spray-dried sap) (GHA I 59)	2.2	196.00 ±53.09	190.27 ±27.78	190.27 ±29.54	192.60 ±29.09	194.73 ±29.68	196.97 ±29.04	198.60 ±30.18
07	Sample 2 (Spray-dried sap) (GHA I 59)	8.8	231.30 ±61.91	177.27 ±24.48	178.17 ±23.79	180.67 ±25.04	182.03 ±25.31	185.10 ±24.60	189.73 ±23.58
80	Spray-dried sap (GHA I 59)	35	167.48 ±36.75	164.90 ±22.54	166.63 ±23.08	168.43 ±22.66	171.67 ±24.42	174.90 ±25.70	178.57 ±23.58
60	Elga Option 4 purified water (control)		165.50 ±49.27	193.73 ±22.37	196.87 ±21.86	198.07 ±21.02	199.83 ±20.21	204.93 ±18.65	207.13 ±18.22



# TABLE 1: HISTOLOGICAL EVALUATION OF LIVER SECTIONS FROM MALE RATS Sample 1

GROUP	1: 100 mg/k	g Sample 1		GROUP	2: 400 mg/k	g Sample 1	
	Animal no	Hepatic lesio	ns		Animal no	Hepatic lesions	
Day 7	01	NPL			07		FHS1+
	02	NPL			08	NPL C1+	
	03	NPL C1+			09	NPL	
Day 13	04	NPL	MLC	Day 13	10		DHS1+
	05		FHS1+		11	NPL	
	06	NPL			12		DHS1+

GROUP :	3: 1 600 mg	/kg Sample 1	GROUP	4: 3 200 mg	J/kg Sample 1	
	Animal no	Hepatic lesions		Animal no	Hepatic lesions	
Day 7	13	NPL		19	NPL	
	14.	NPL		20	NPL	
	15	NPL		21	NPL.	
Day 13	16	NPL	Day 13	22		DHS1+
	17	DHS1+		23		FHS1+
	18	NPL		24	NPL	••

# GROUP 5: CONTROL: ELGA OPTION 4 PURIFIED WATER: RESTRICTED FOOD INTAKE

GROUP !	5: Control: E	lga option 4 purified water
	Animal no	Hepatic lesions
Day 7	25	NPL MLC
	26	NPL
	27	NPL
Day 13	28	DHS1+
	29	DHS1+
	30	NPL

# Legend:

С	=	Congestion
DHS	=	Diffuse hydropic cell swelling
FHS	=	Focal hydropic cell swelling
NPL	=	No parenchymal lesions
MLC	=	Minimal lymphocytic cuffing
1+	=	mild
2+	=	moderate
3+	=	severe

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# TABLE 2: HISTOLOGICAL EVALUATION OF LIVER SECTIONS FROM MALE RATS Sample 2

GROUP	6: 2.2 mg/kg	Sample 2			GROUP	7: 8.8 mg/k	g Sample 2	
	Animal no	Hepatic le	sions			Animal no	Hepatic lesions	
Day 7	31	NPL				37	NPL	
	32	NPL	MLC			38	NPL	
	33			FHS1+		39	NPL C1+	
Day 13	34	NPL			Day 13	40		DHS1+
	35			DHS1+		41	NPL	
	36	NPL				42	MLC	FHS1+

GROUP !	3: 35 mg/kg	Sample 2
	Animal no	Hepatic lesions
Day 7	43	NPL
	44	NPL
	45	NPL
Day 13	46	NPL
	47	NPL C1+
	48	MLC FHS1+

# GROUP 9: CONTROL: ELGA OPTION 4 PURIFIED WATER

9: Control: E	lga option 4 purified water
Animal no	Hepatic lesions
49	NPL
50	NPL
51	FHS1+
52	DHS1+
53	NPL
54	FHS1 +
	Animal no 49 50 51 52 53

# Legend:

C = Congestion

DHS = Diffuse hydropic cell swelling

FHS = Focal hydropic cell swelling

NPL = No parenchymal lesions

MLC = Minimal lymphocytic cuffing

1 + = mild

2+ = moderate

3 + = .severe

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No specific lesions were recorded in the liver sections from the experimental rats which received the frozen sap as well as the spray-dried sap that could be attributed to the oral adminstration of the abovementioned chemicals. The hydropic cell swelling recorded in both control and experimental rats may indicate normal metabolic cell swelling and anoxic changes. Minimal foci of lymphocytic perivascular cuffing were found in some animals and is most likely an incidental observation. In a few rats congestion of mild degree is present in the hepatic sinusoids and should be regarded as an incidental observation.

\* \* \* \* \* \* \* \* \*

An important feature of the invention shown by the results of this study is that no tolerance to any of the samples developed over the test period. This may provide considerable benefit, particularly in relation to the use of the compounds and compositions of the invention in the treatment of obesity.

While the compounds and compositions of the invention have primarily been described in relation to their properties as appetite suppressants, it should be noted that this expression – "appetite suppressant" – is used herein to denote activity which tends to limit appetite and/or increase the sense of satiety, and thus tends to reduce total calorific intake; this in turn tends to counteract obesity. Accordingly, this invention extends to a method of treating, preventing or combating obesity in a human or non-human animal which comprises administering to said human or non-human animal an obesity treating, preventing or combating amount of a compound of formula (2). A preferred embodiment of this aspect of the invention utilises a composition or extract containing a compound of formula (1).

The term "animal" as used herein extends to, but is not restricted to, companion animals, e.g. household pets and domesticated animals; non-limiting examples of such animals include cattle, sheep, ferrets, swine, camels, horses, poultry, fish, rabbits, goats, dogs and cats.

As an anorectic agent or in the treatment or prevention of obesity in a human, a compound of formula (2), preferably of formula (1), or the composition defined in any one of claims 9 and 25-31 hereafter, is advantageously administered to said human in a dosage amount of from about 0.01 mg/kg/day to about 10 mg/kg/day. A preferred dosage range is 0.05 mg/kg/day to 0.5 mg/kg/day. When using the spray dried powder form of the extract of this invention, a preferred dosage range is 0.1 mg/kg/day to 20 mg/kg/day; especially preferred is 0.5 mg/kg/day to 5 mg/kg/day.

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